

Towards a sustainable future

Study of new technologies
for **wastes management** and
CO₂ emissions reduction

Azzurra Zucchini, Paola Comodi



A.D. 1308
unipg

UNIVERSITÀ DEGLI STUDI
DI PERUGIA

A.D. 1308
unipg

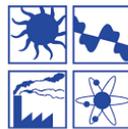
DIPARTIMENTO
DI FISICA E GEOLOGIA

Dipartimenti:

- Ingegneria
- Scienze Farmaceutiche
- Chimica, Biologia e Biotecnologie
- Scienze agrarie, alimentari e ambientali

In collaboration with:

COLACEM
forte • sostenibile



CIRIAC
Centro Interuniversitario
di Ricerca sull'Inquinamen
da Agenti Fisici - "Mauro F

Partially founded by:



Fondazione Cassa di Risparmio di Perugia



European
Commission

European Regional Development Fund 2014-2020
(ERDF 2014-2020)

European Social Fund 2007-2013 (ESF 2007-2013)



The Research Team...past and present



Dr. Azzurra Zucchini



Prof. Paola Comodi



Prof. Costanza Cambi



Prof. Manuela Cecconi



Prof. Franco Cotana



Prof. Gianluca Cavalaglio



Dr. Maximiliano Fastelli



Prof. Francesco Frondini



Prof. Riccardo Vivani



Prof. Paola Sassi



Dr. Alessandro Michele



Prof. Paolo Blasi



Dr. Francesco Vetere



Dr. Alessandro Neri, Dr. Fabio Santinelli.



Luca Bartolucci



Gianluca Polidori

BSc, MSc and PhD students:

- **Dr. Edoardo Rampini**
- **Dr. Giada Santarelli**
- **Dr. Giulia Guidobaldi**
- **Dr. Silvia Gentili**
- **Dr. Serena Casagrande**



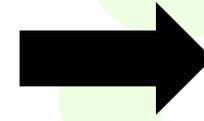
#1 The Circular Economy

- ✓ To reduce the consumption of natural raw materials
- ✓ To protect human health and the environment

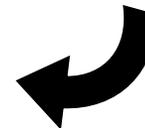
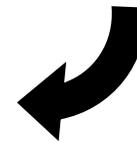
INDUSTRIAL MINERALS SECTOR CONTRIBUTION TO CIRCULAR ECONOMY



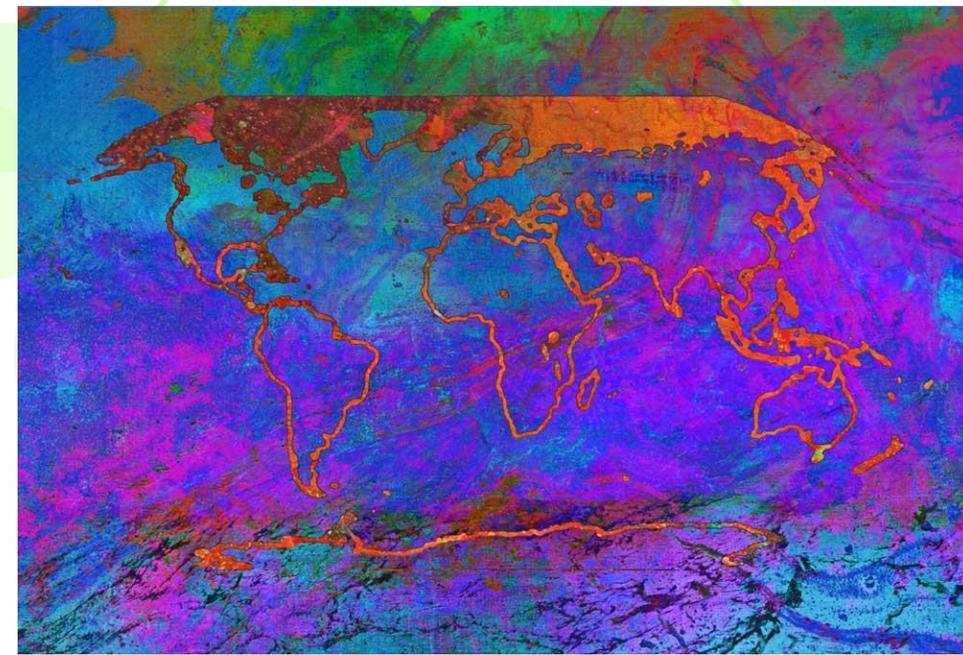
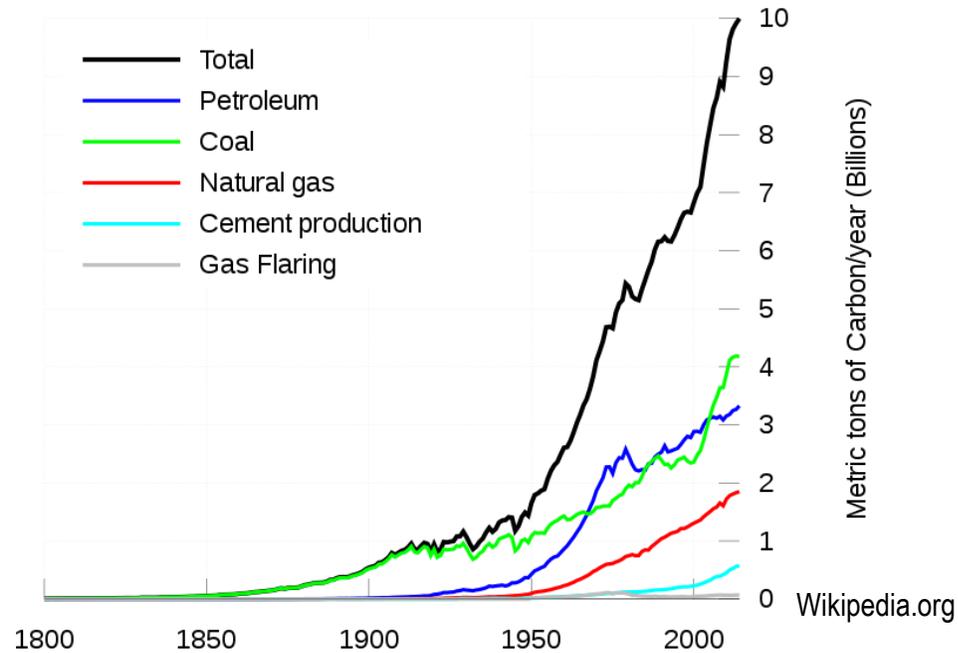
Protagonisti:
"geomateriali"



Parlamento Europeo 2015



#2 The anthropic CO₂ emissions



Changing by Alisa Singer
"As we witness our planet transforming around us we watch, listen, measure ... respond."
www.environmentalgraphiti.org – 2021 Alisa Singer.

A.1 It is unequivocal that human influence has warmed the atmosphere, ocean and land. Widespread and rapid changes in the atmosphere, ocean, cryosphere and biosphere have occurred.

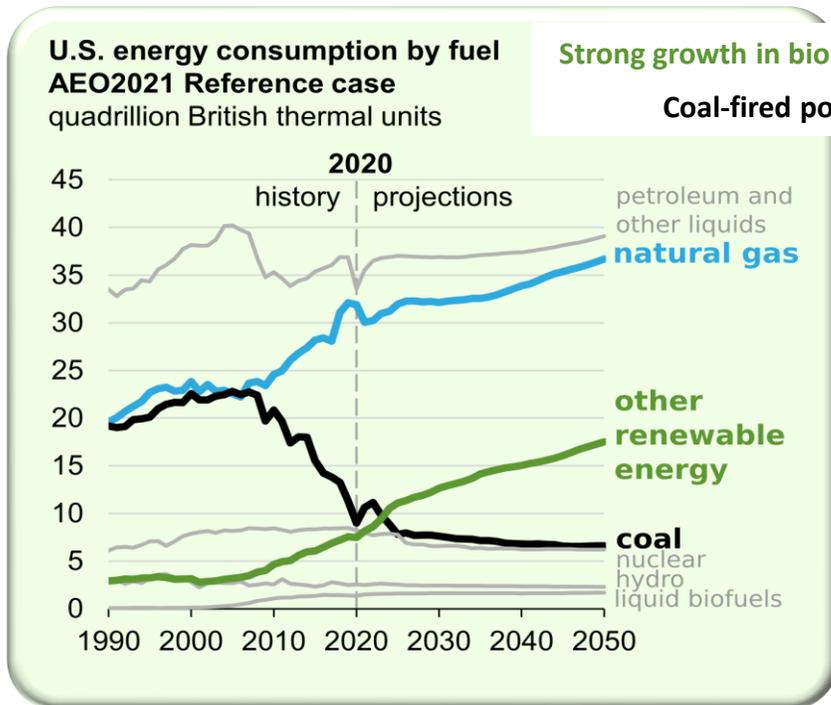
“Observed increases in well-mixed greenhouse gas (GHG) concentrations since around 1750 are unequivocally caused by human activities. Since 2011 ... concentrations have continued to increase in the atmosphere, reaching annual averages of 410 ppm for carbon dioxide (CO₂), 1866 ppb for methane (CH₄), and 332 ppb for nitrous oxide (N₂O) in 2019. Land and ocean have taken up a near-constant proportion (globally about 56% per year) of CO₂ emissions from human activities over the past six decades...”.



Fly ashes and their employment in civil engineering and agriculture

Biomass fly ashes: amount of solid, inorganic residue left after the complete burning of biomasses.

(James et al. 2012 DOI: 10.3390/en5103856, Torquati et al. 2016 DOI:10.1016/j.biombioe.2016.09.017, Cavalaglio et al. 2020 DOI: 10.3390/su12166678)



Strong growth in biomass plants → production of large quantities of ash
Coal-fired power plants → undergoing decommissioning



Geotechnical engineering, industrial processes and agriculture: sometimes problems in raw materials supplying

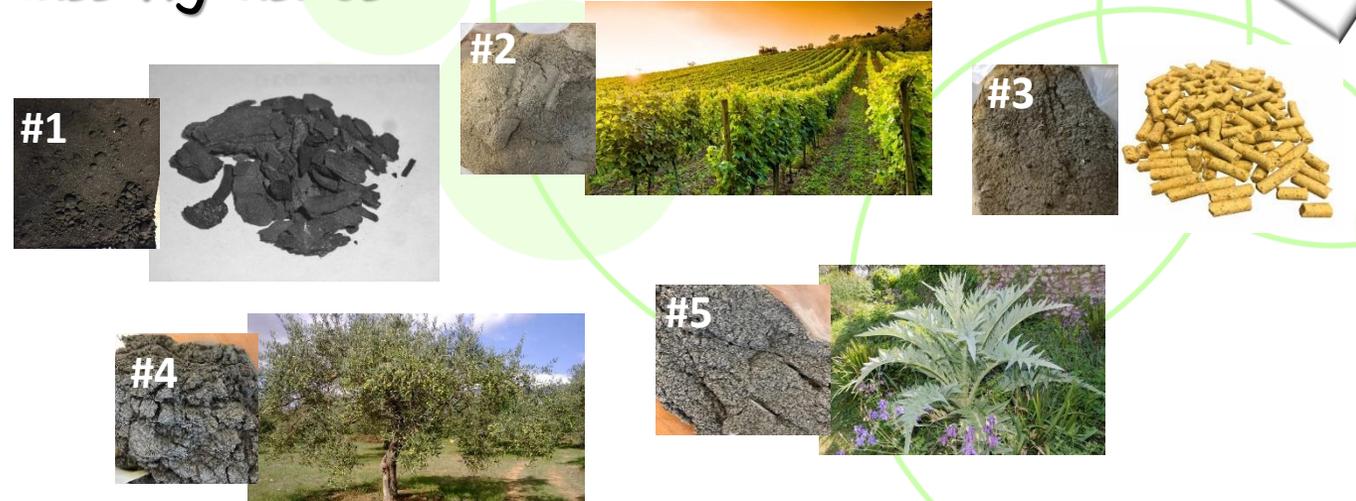
Modified from EIS's Annual Energy Outlook 2021

Fly ashes and their employment in civil engineering and agriculture

The Circular Economy

Biomass fly ashes

Code	Origin
#1	Gasification power plant, Magione (PG) Italy
#2	Grapevine prunings, Torgiano (PG) Italy
#3	Wood pellet
#4	Olive tree prunings
#5	Cardoon plants



Analytical techniques

TG-DTA
@UNIPG

FE-SEM
@UNIPG

XRPD
@UNIPG

DLS
@UNIPG

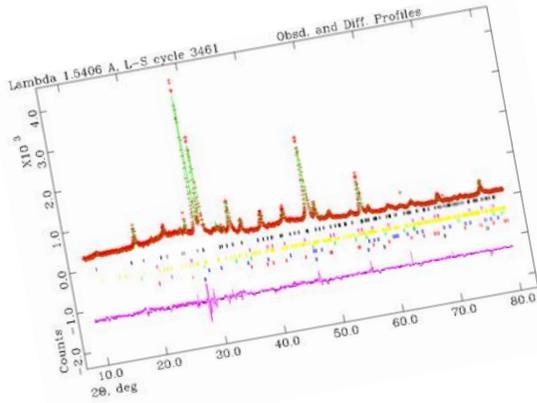
PAHs
@ARPA Umbria

EMPA
@UNIMI

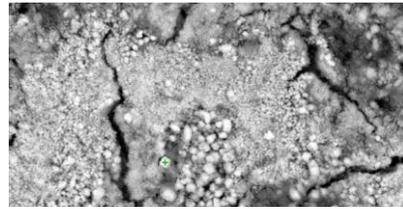
LA-ICP-MS
@UNIPG

Fly ashes and their employment in civil engineering and agriculture

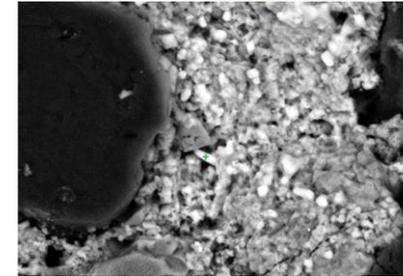
Results



Apatite and sylvite
Small dimensions

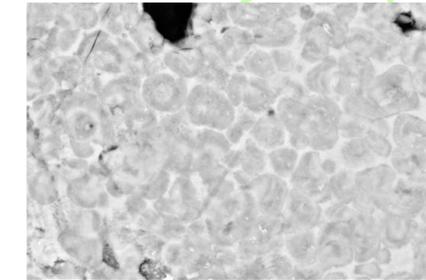


Apatite crystals
roundly shaped at sub-micrometric to deci-micrometric sizes



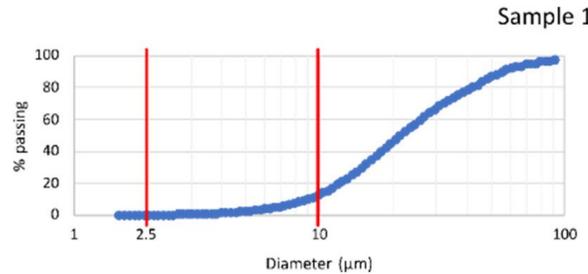
Elongated K_2O and KCl crystals

Comodi et al. (2021) 10.3390/su13116052



Devitrification structures

High P and K content.



Red lines: positions of the 2.5 and 10 µm diameter particles. They identify the **inhalation properties**.

- ▶ Less than 1 vol% of particles with dimensions under 2.5 µm .
- ▶ Particles with diameters less than 10 µm vary from 5 to 40 vol%.

Large reaction surface, one the most important parameters for **bioavailability**.
Good fertilizer for agricultural lands.

High Ca content
Supposedly stored in the amorphous phase



Biomass fly ashes could be used in combination with or as **substitutions for traditional binders** (e.g. CaO) in soil stabilization.

Good amendments in soils, usually clayey soils, with low geotechnical properties

Fly ashes and their employment in civil engineering and agriculture

To move on...towards Circular Economy

1. Addition of Ca from biomass fly ashes in soils with poor geotechnical properties.



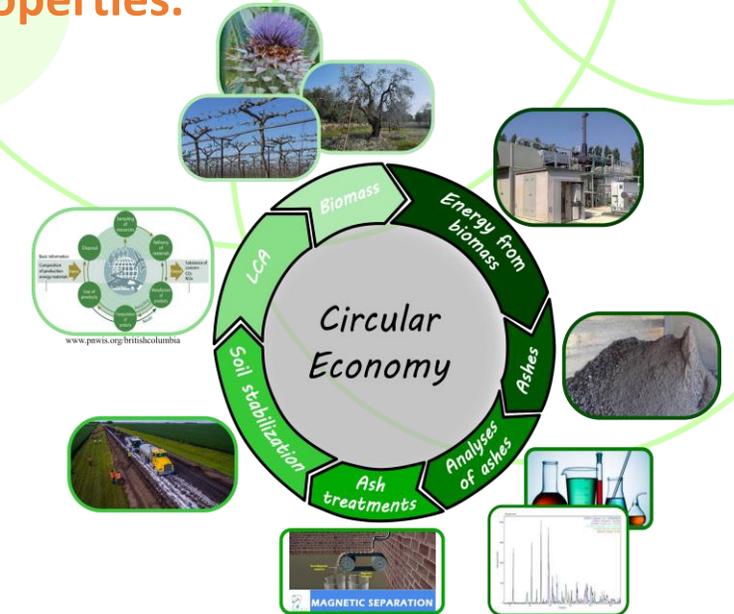
2. Biomass fly ashes used as agricultural fertilizers.



The above-mentioned scenarios fit well with the principles of the

Circular Economy

Waste materials, such as biomass fly ashes, are recycled and used as second generation products.



Collocazione PTSR 2021-2023:
Ambito di ricerca nuovo: 1
«Earth System and Global Changes»

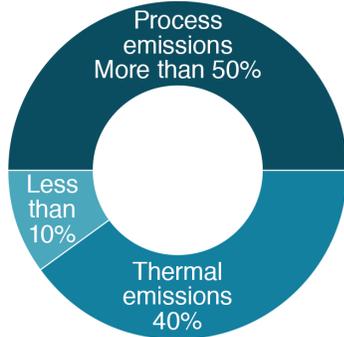
The Circular Economy

Nanomaterials: a way to decrease the CO₂ emissions in cement industry

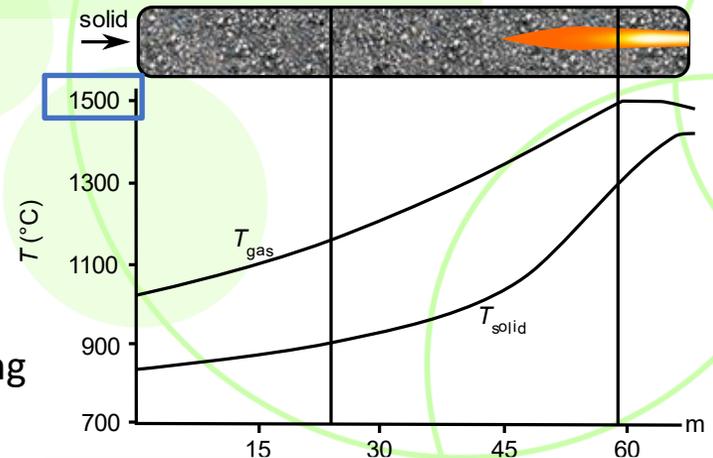
CO₂ emissions reduction

The production of “clinker” accounts for most of the CO₂ emissions of cement production

- Quarrying & transport
- Grinding & preparation of raw materials
- Cooling, grinding, mixing



Clinker production → rotary kilns feeding
 → decarbonation during firing



Source: Chatham House

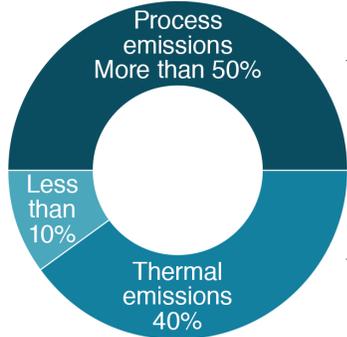
BBC

Nanomaterials: a way to decrease the CO₂ emissions in cement industry

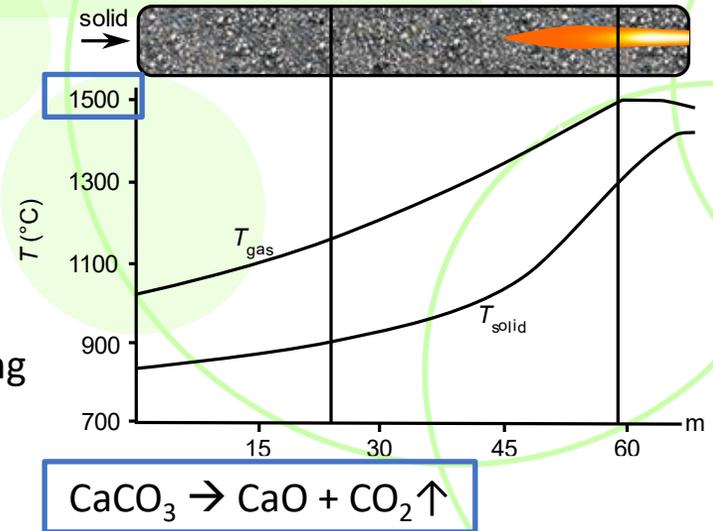
CO₂ emissions reduction

The production of “clinker” accounts for most of the CO₂ emissions of cement production

- Quarrying & transport
- Grinding & preparation of raw materials
- Cooling, grinding, mixing



Clinker production → rotary kilns feeding
 → decarbonation during firing



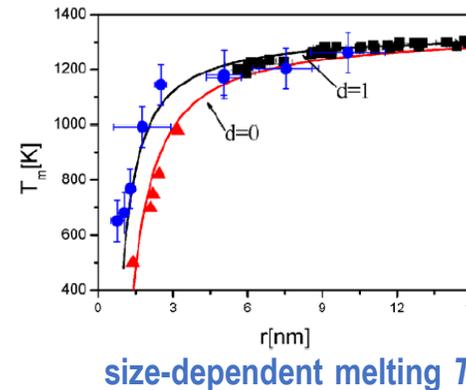
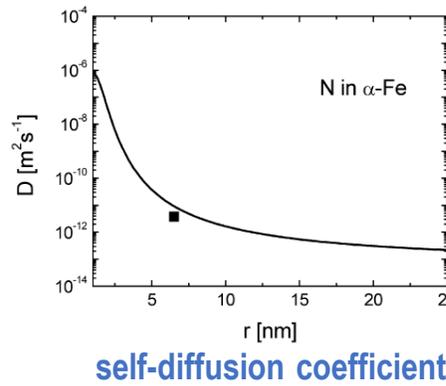
Source: Chatham House

BBC

Ca²⁺ diffusion is the driven mechanism for clinker minerals formation



Telschow (2012) Clinker Burning Kinetics and Mechanism - PhD thesis



Dimension to nm

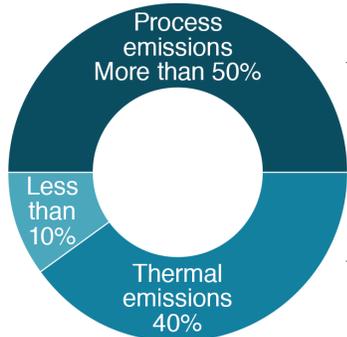
Melting T
 ↓
 ↑
 Diffusion coefficient

Cohn (1948) DOI: 10.1021/cr60133a002
 Jiang et al. (2004) DOI: 10.1016/j.ssc.2004.03.033

Nanomaterials: a way to decrease the CO₂ emissions in cement industry

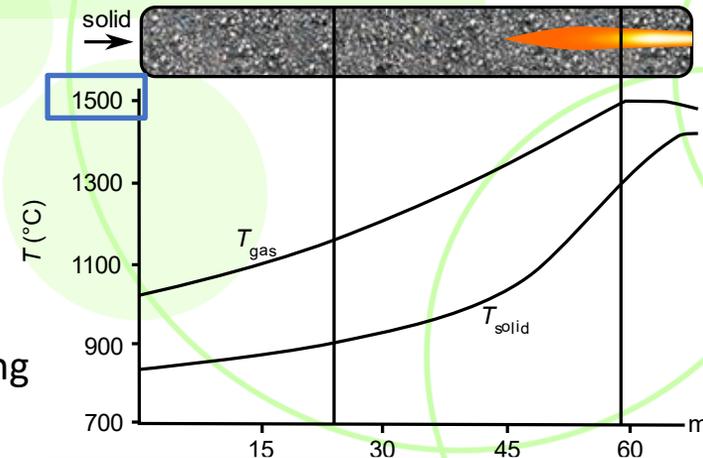
The production of “clinker” accounts for most of the CO₂ emissions of cement production

- Quarrying & transport
- Grinding & preparation of raw materials
- Cooling, grinding, mixing



Clinker production

- rotary kilns feeding
- decarbonation during firing

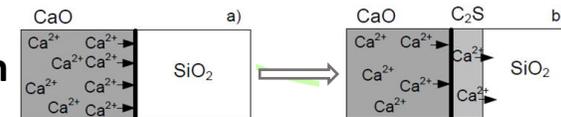


CO₂ emissions reduction

Source: Chatham House

BBC

Ca²⁺ diffusion is the driven mechanism for clinker minerals formation



Telschow (2012) Clinker Burning Kinetics and Mechanism - PhD thesis



Limestone + Clay + Nano-materials

Nano-CaO ?

Nanomaterials: a way to decrease the CO₂ emissions in cement industry

Clinker production in Lab



CO₂ emissions reduction

Analytical techniques

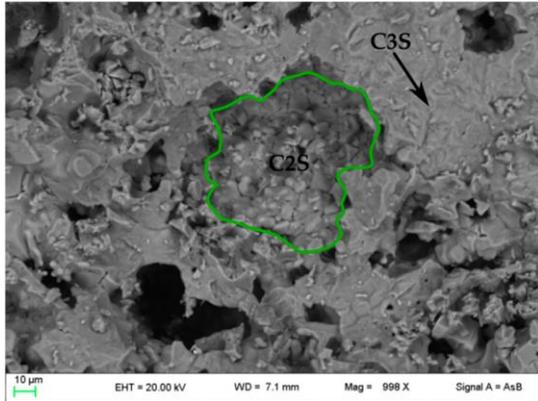


Nanomaterials: a way to decrease the CO₂ emissions in cement industry

CO₂ emissions reduction

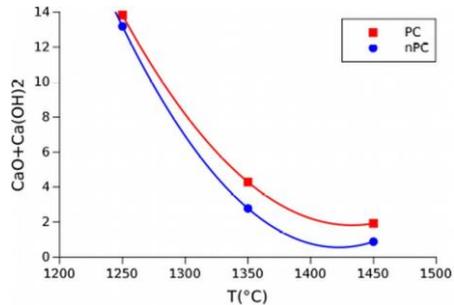
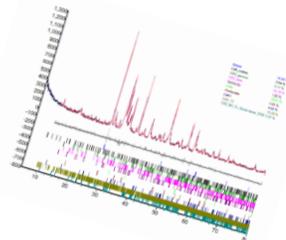
Results

Zucchini et al (2019) DOI: 10.3390/ma12111787



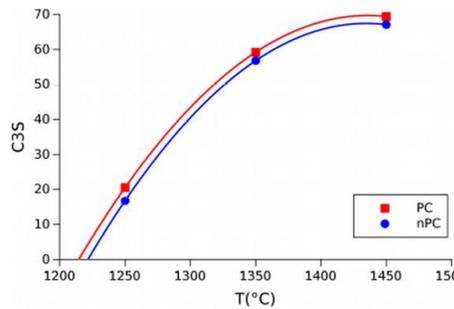
Belite nest @ 1350°C with nano-CaO

Higher **crystallinity** of the nano-added clinker

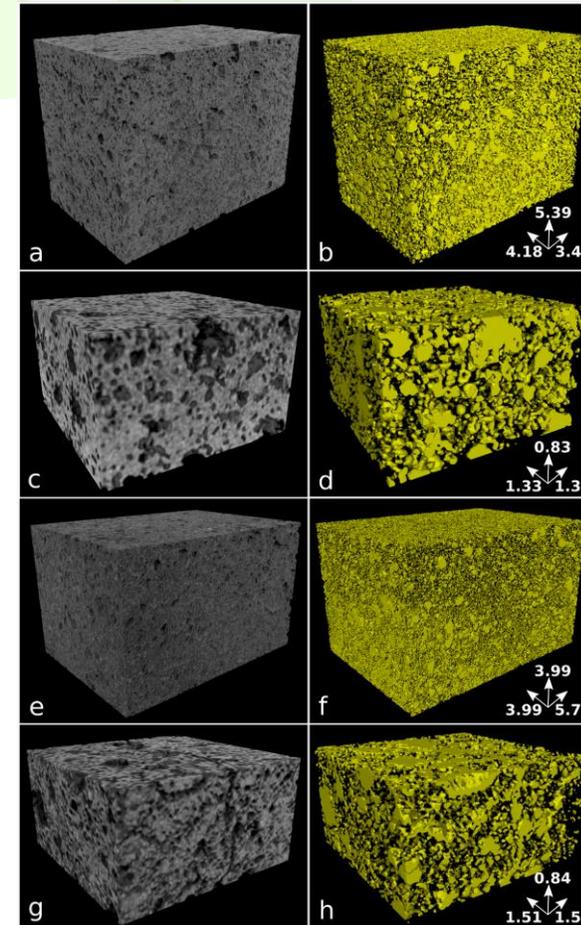
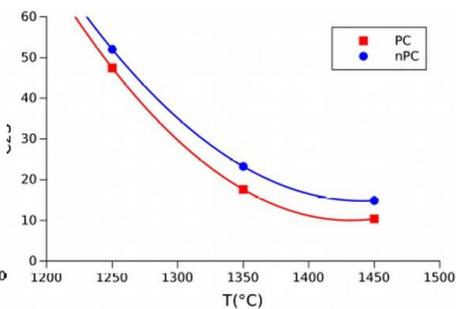


Decrease of free lime

Improvement in the nano-added clinker **cooking efficiency**



Higher **reactivity** of the nano-added clinkers



1450°C with nano-CaO

1450°C

1350°C with nano-CaO

1350°C

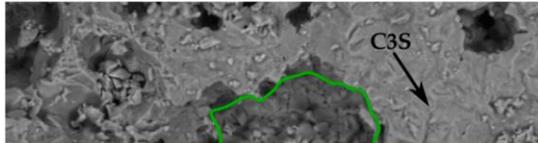
Decrease in the total **porosity** of the clinker prepared by using nano-CaO

Nanomaterials: a way to decrease the CO₂ emissions in cement industry

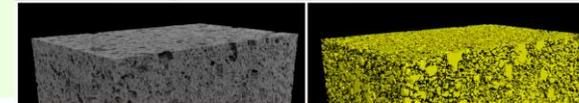
CO₂ emissions reduction

Results

Zucchini et al (2019) DOI: 10.3390/ma12111787



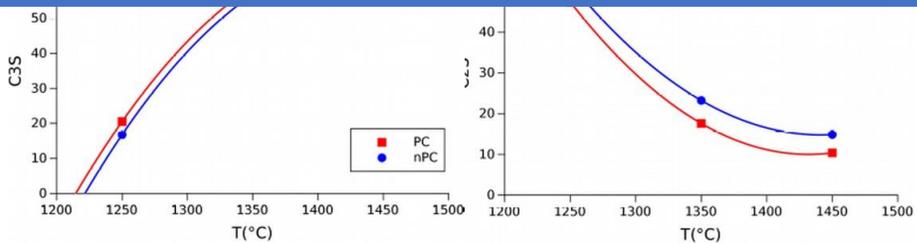
Belite nest @ 1350°C with nano-CaO



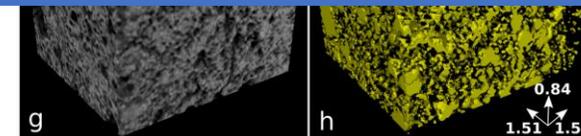
Collaborative actions
COLACEM
forte • sostenibile

Increase of the nano-added clinkers cooking efficiency

- ➔ Decrease of the cooking T of approximately 100°C means **low energy requirement for the minerals formation reactions** during Portland clinker production.
- ➔ **Decrease of the CO₂ emissions** from both the **pre-treatment of starting materials** and from the **cooking process**
- ➔ **Increase of the plant productivity**



Higher reactivity of the nano-added clinkers



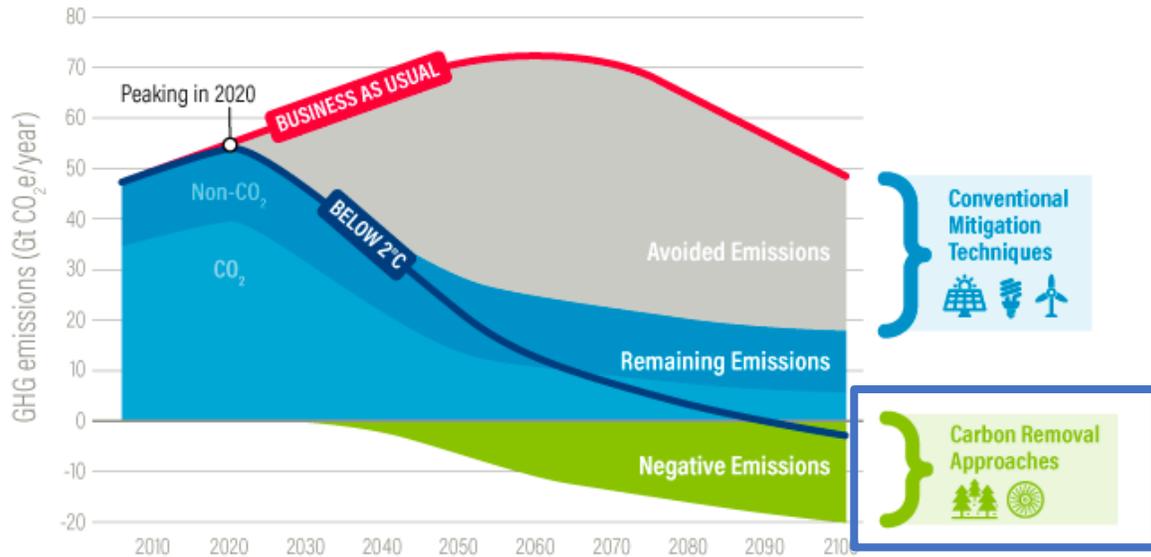
1350°C

Decrease in the total porosity of the clinker prepared by using nano-CaO

Nanomaterials: a way to increase the kinetics of the CO₂ Mineralization

To move on...towards *Negative emissions*

CO₂ emissions reduction



Ex-situ CO₂ Mineralization Reactions (CMRs)
 Reaction between CO₂ and natural silicate minerals (pyroxenes, olivine and serpentine) producing at least one carbonate type.

$$(Mg,Ca)_xSi_yO_{x+2y+z}H_{2z(s)} + xCO_{2(g)} \rightarrow x(Mg,Ca)CO_{3(s)} + ySiO_{2(s)} + zH_2O_{(l)}$$


Slow reaction kinetics

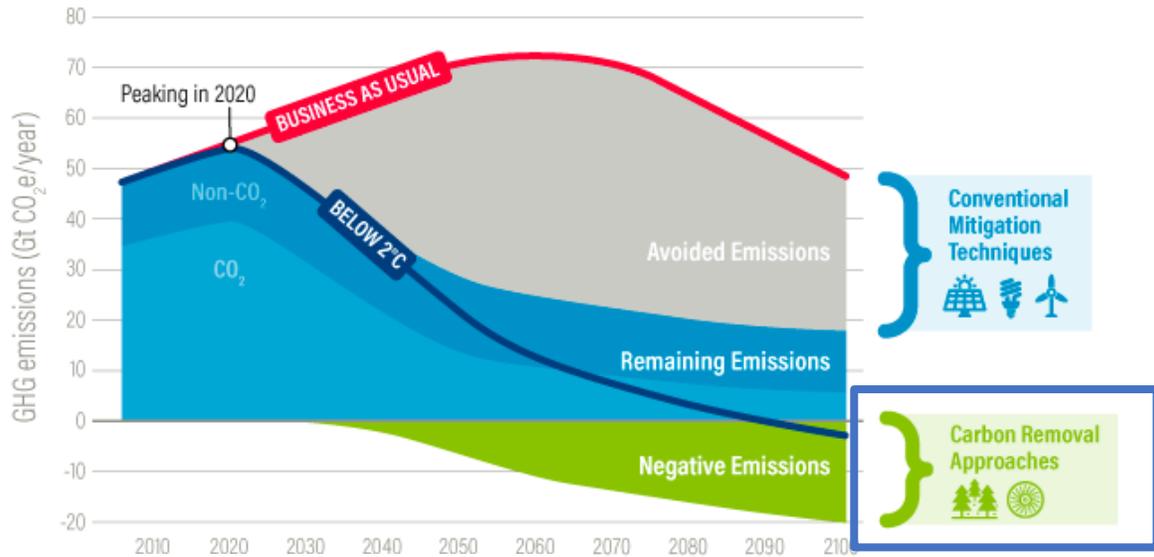
Source: Adapted from UNEP 2016.
 For more information, visit wri.org/carbonremoval.

WORLD RESOURCES INSTITUTE

Nanomaterials: a way to increase the kinetics of the CO₂ Mineralization

CO₂ emissions reduction

To move on...towards *Negative emissions*

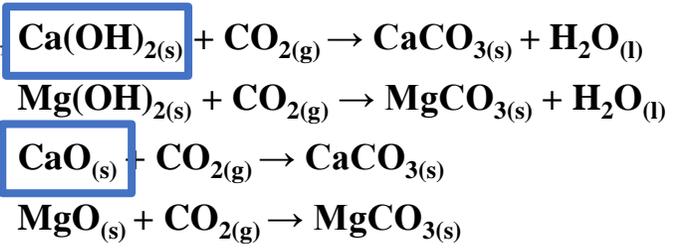
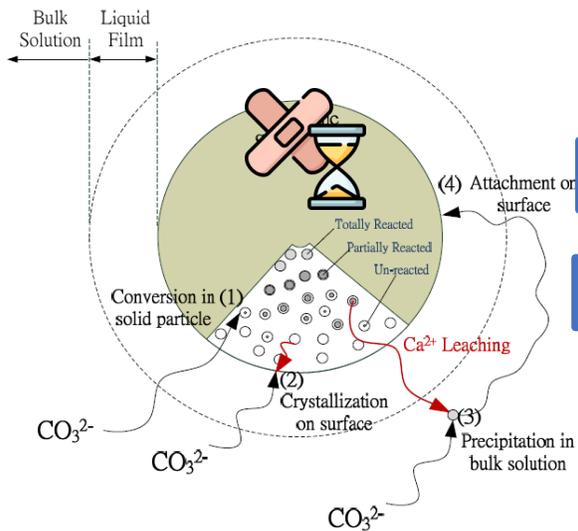


Ex-situ CO₂ Mineralization Reactions (CMRs)
 Reaction between CO₂ and natural silicate minerals (pyroxenes, olivine and serpentine) producing at least one carbonate type.
 $(Mg,Ca)_xSi_yO_{x+2y+z}H_{2z(s)} + xCO_{2(g)} \rightarrow x(Mg,Ca)CO_{3(s)} + ySiO_{2(s)} + zH_2O_{(l)}$



Slow reaction kinetics

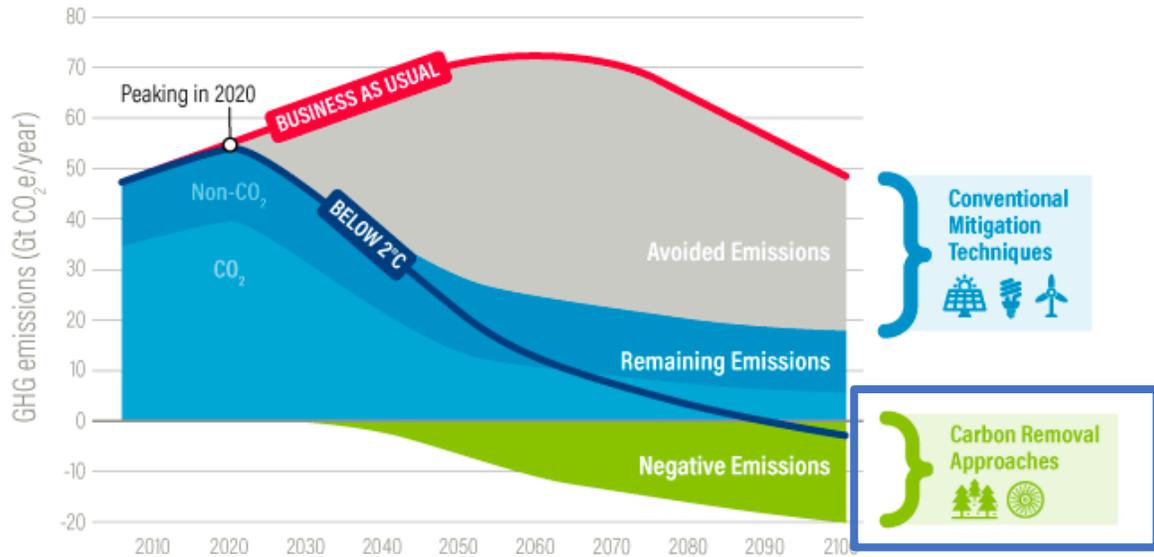
WORLD RESOURCES INSTITUTE



Nanomaterials: a way to increase the kinetics of the CO₂ Mineralization

CO₂ emissions reduction

To move on...towards *Negative emissions*

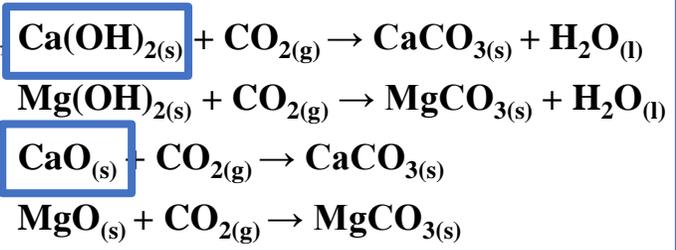
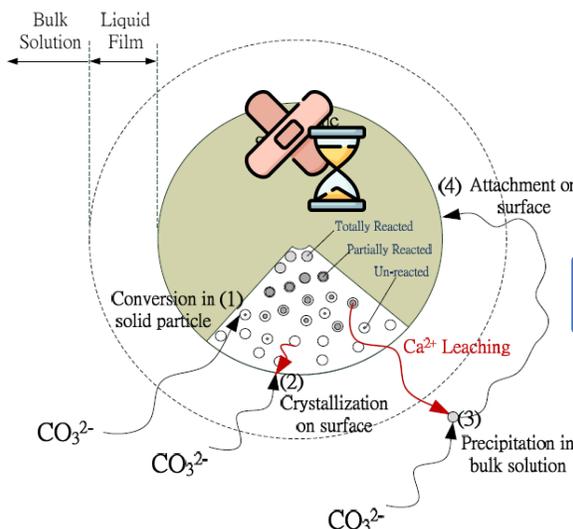


Ex-situ CO₂ Mineralization Reactions (CMRs)
 Reaction between CO₂ and natural silicate minerals (pyroxenes, olivine and serpentine) producing at least one carbonate type.
 $(Mg,Ca)_xSi_yO_{x+2y+z}H_{2z(s)} + xCO_{2(g)} \rightarrow x(Mg,Ca)CO_{3(s)} + ySiO_{2(s)} + zH_2O_{(l)}$



Slow reaction kinetics

WORLD RESOURCES INSTITUTE



Main mechanism that can affect the rate and extent of carbonation:

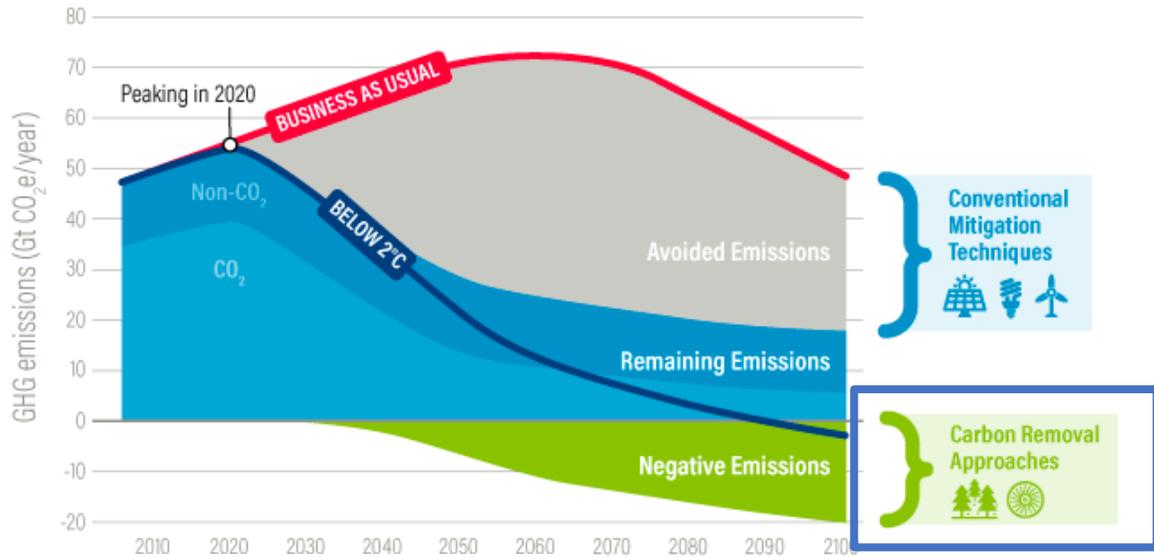
- transportation-controlled mechanisms such as CO₂ and Ca²⁺-ions diffusion to/from reaction sites;

Pan et al. (2012) DOI: 10.4209/aaqr.2012.06.0149

Nanomaterials: a way to increase the kinetics of the CO₂ Mineralization

CO₂ emissions reduction

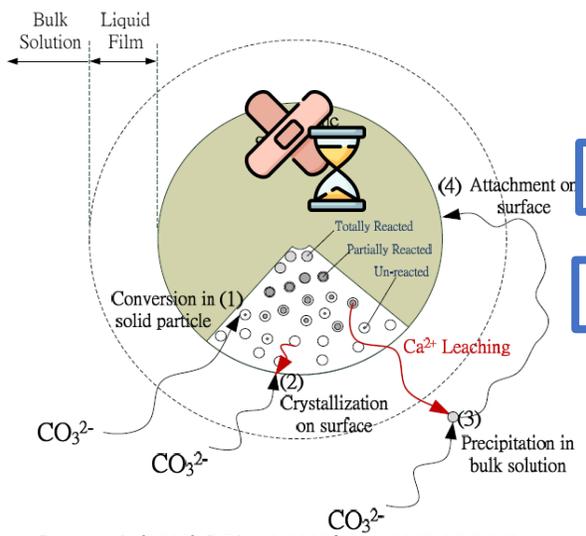
To move on...towards *Negative emissions*



Ex-situ CO₂ Mineralization Reactions (CMRs)
 Reaction between CO₂ and natural silicate minerals (pyroxenes, olivine and serpentine) producing at least one carbonate type.
 $(Mg,Ca)_xSi_yO_{x+2y+z}H_{2z(s)} + xCO_{2(g)} \rightarrow x(Mg,Ca)CO_{3(s)} + ySiO_{2(s)} + zH_2O_{(l)}$



Slow reaction kinetics



WORLD RESOURCES INSTITUTE

Ca(OH)_{2(s)}
 Mg(OH)_{2(s)}
 CaO(s) + CO_{2(l)}
 MgO(s) + CO_{2(l)}

Nano-CaO ?

Main mechanism that can affect the rate and extent of carbonation:

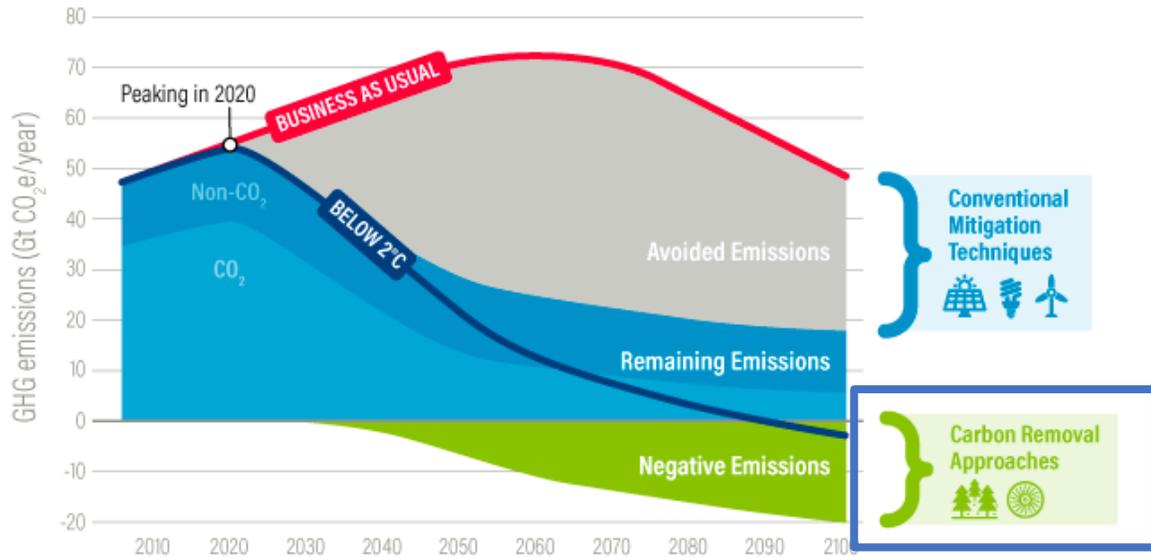
- transportation-controlled mechanisms such as CO₂ and Ca²⁺-ions diffusion to/from reaction sites;

Pan et al. (2012) DOI: 10.4209/aaqr.2012.06.0149

Nanomaterials: a way to increase the kinetics of the CO₂ Mineralization

To move on...towards *Negative emissions*

CO₂ emissions reduction



Ex-situ CO₂ Mineralization Reactions (CMRs)

Reaction between CO₂ and natural silicate minerals (pyroxenes, olivine and serpentine) producing at least one carbonate type.

$$(Mg,Ca)_xSi_yO_{x+2y+z}H_{2z(s)} + xCO_{2(g)} \rightarrow x(Mg,Ca)CO_{3(s)} + ySiO_{2(s)} + zH_2O_{(l)}$$


Slow reaction kinetics

Collocazione PTSR 2021-2023:
Ambito di ricerca già attivato: 5
«Nanoscienze»
Ambito di ricerca nuovo: 1
«Earth System and Global Changes»

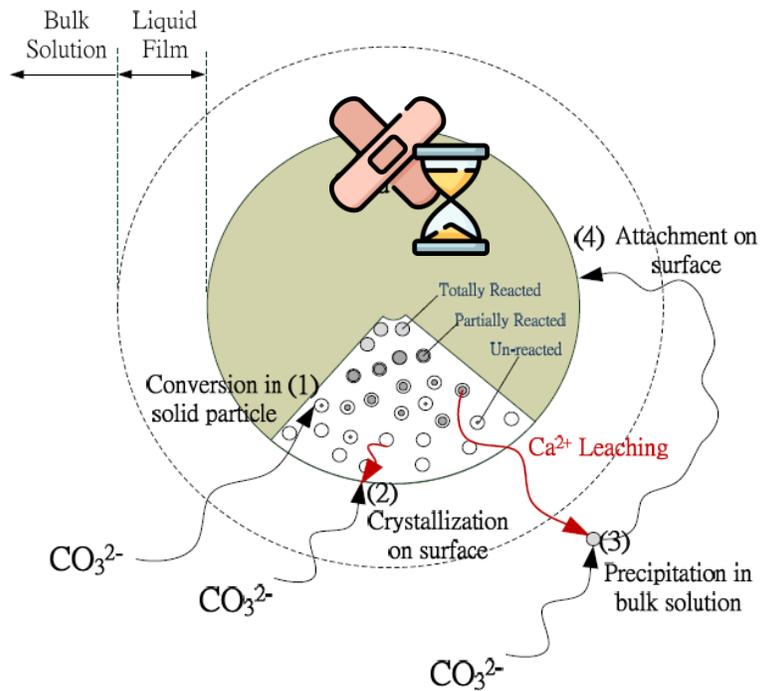
Source: Adapted from UNEP 2016.
For more information, visit wri.org/carbonremoval.



Use of nano-materials to improve the kinetics of the CO₂ mineralization reactions.

The new IDEA

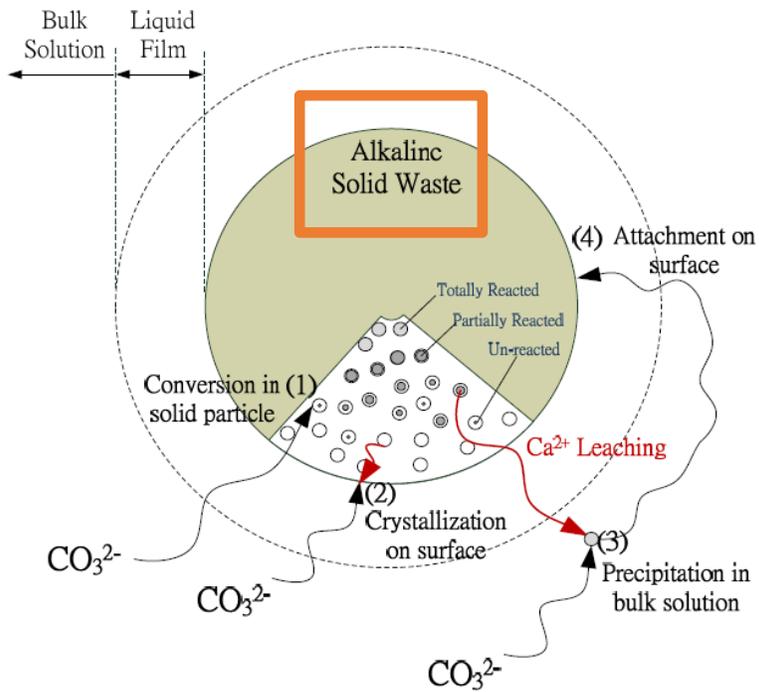
Joining Circular Economy and Negative Emissions



Pan et al. (2012) DOI: 10.4209/aaqr.2012.06.0149

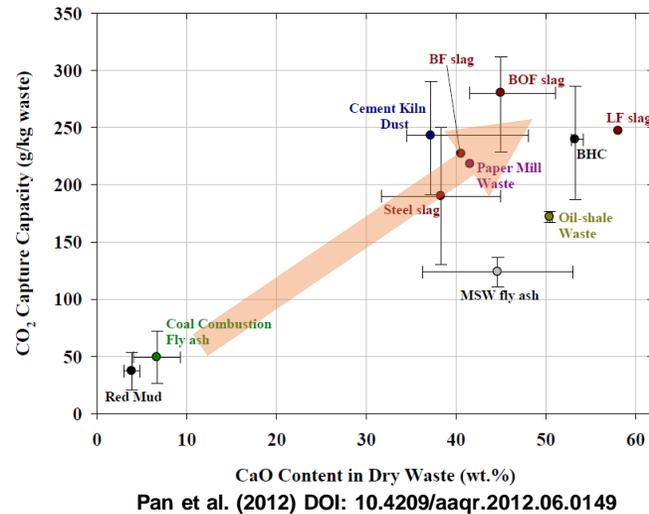
Joining Circular Economy and Negative Emissions

Ex situ CO₂ mineralization reactions by means of waste materials and nanomaterials



Pan et al. (2012) DOI: 10.4209/aaqr.2012.06.0149

Substitution of natural minerals with alkaline wastes



▶ Case study #1 Nano-CaO



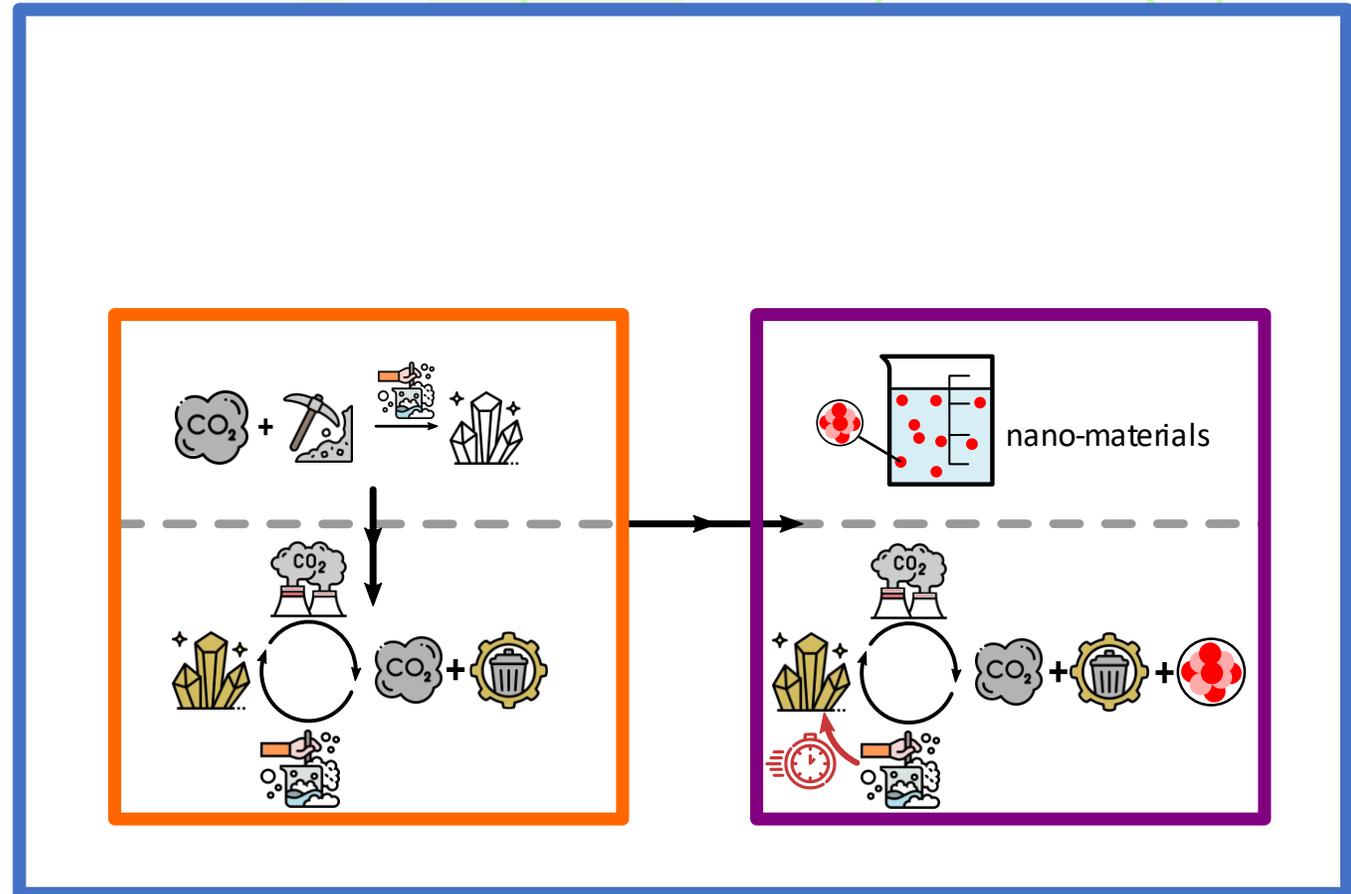
▶ Case study #2 Biomass fly-ashes Being enriched in CaO.

▶ Joining #1 and #2

Joining Circular Economy and Negative Emissions

Ex situ CO₂ mineralization reactions by means of **waste materials** and **nanomaterials**

2. Reagents at **low cost** that will also meet the status of “**end of waste**”
3. **Overcoming the slow reaction kinetics**



Joining Circular Economy and Negative Emissions

Ex situ CO₂ mineralization reactions by means of **waste materials** and **nanomaterials**

- 1.** Stable carbonate minerals
- 2.** Reagents at **low cost** that will also meet the status of “**end of waste**”
- 3.** Overcoming the slow reaction kinetics

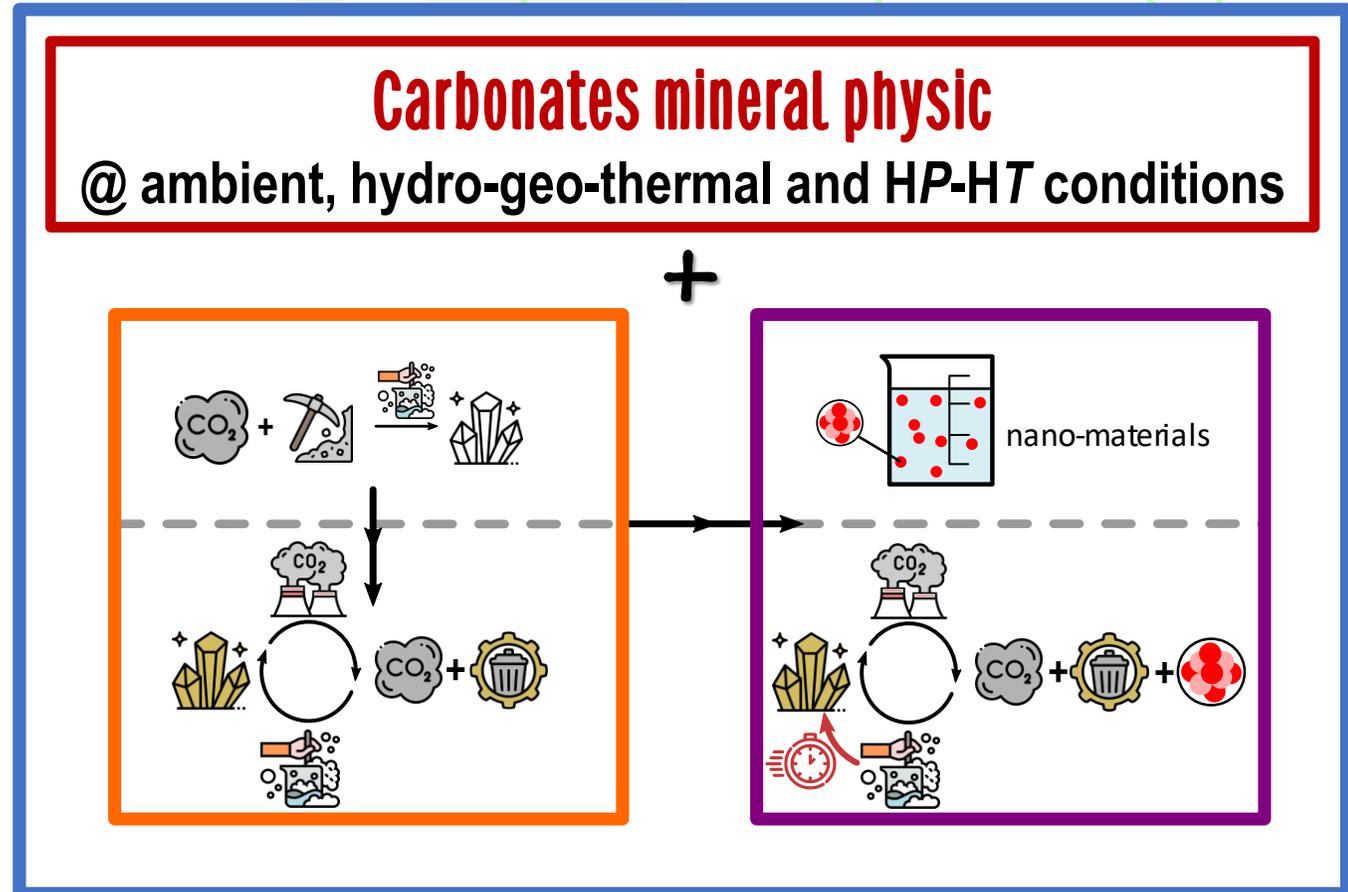
CO₂ mineralization reactions to an advanced state, efficient in terms of

CO₂ capturing reaction kinetics

stability of the final products

in the frame of

circular economy and **sustainability**



Joining Circular Economy and Negative Emissions

Ex situ CO₂ mineralization reactions by means of **waste materials** and **nanomaterials**

- 1.** Stable carbonate minerals
- 2.** Reagents at **low cost** that will also meet the status of “end of waste”
- 3.** Overcoming the slow reaction kinetics

Collocazione PTSR 2021-2023

Ambiti di ricerca già attivati: 5 e 11

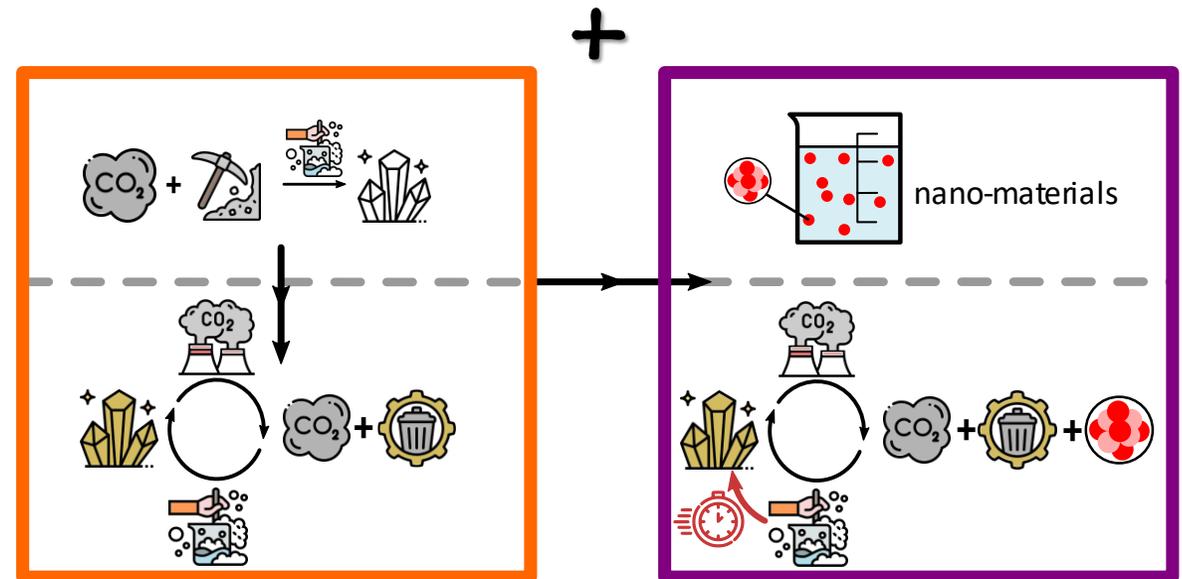
«**Nanoscienze**»

«**Struttura dell'interno della Terra e geodinamica**»

Ambito di ricerca nuovo: 1

«**Earth System and Global Changes**»

Carbonates mineral physic
@ ambient, hydro-geo-thermal and HP-HT conditions



Joining Circular Economy and Negative Emissions

Ex situ CO₂ mineralization reactions by means of **waste materials** and **nanomaterials**

- 1.** Stable carbonate minerals
- 2.** Reagents at **low cost** that will also meet the status of “**end of waste**”
- 3.** Overcoming the slow reaction kinetics

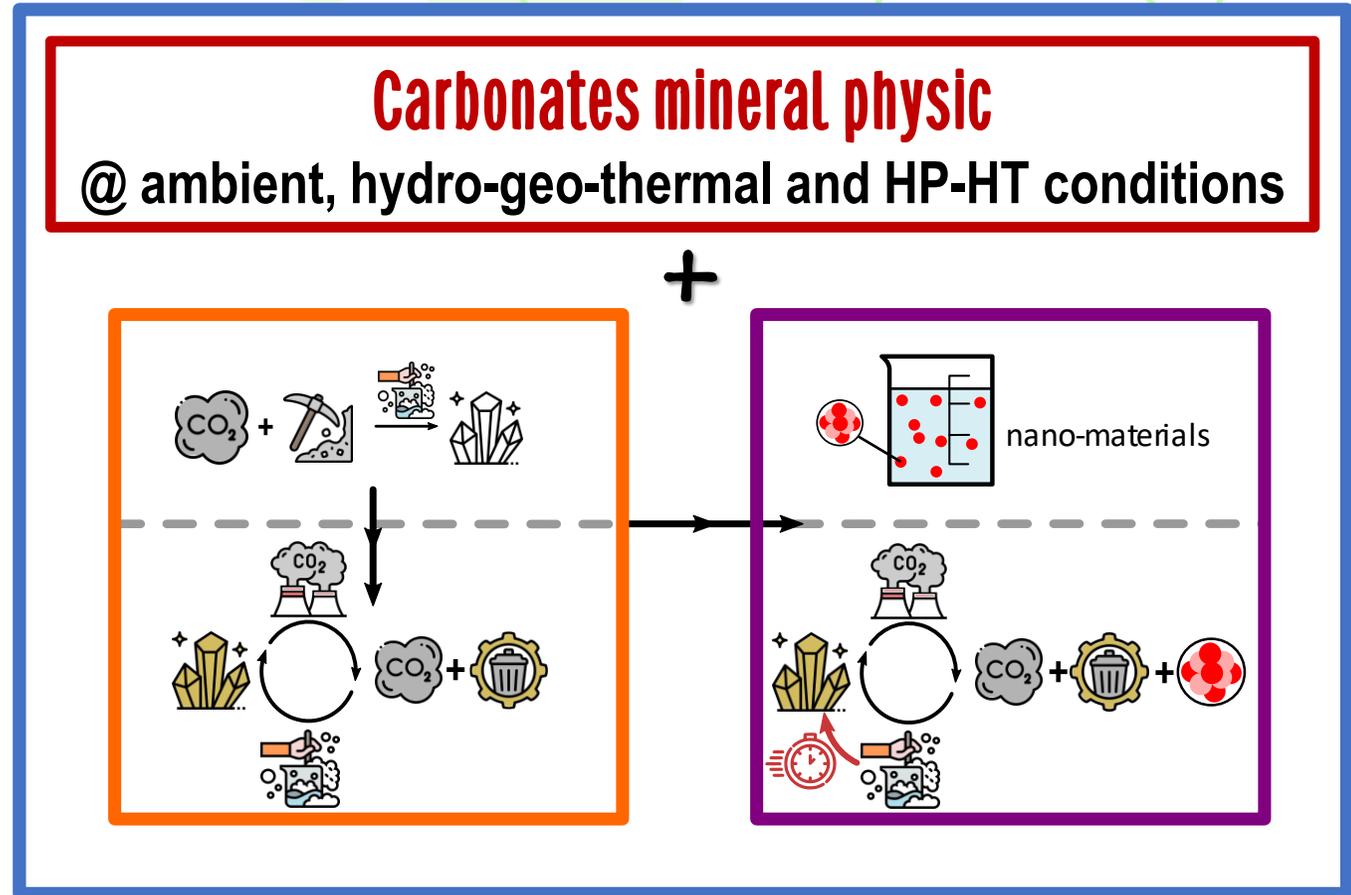
Submitted proposals



European Research Council
Established by the European Commission



Future proposals



Joining Circular Economy and Negative Emissions

Ex situ CO₂ mineralization reactions by means of **waste materials** and **nanomaterials**

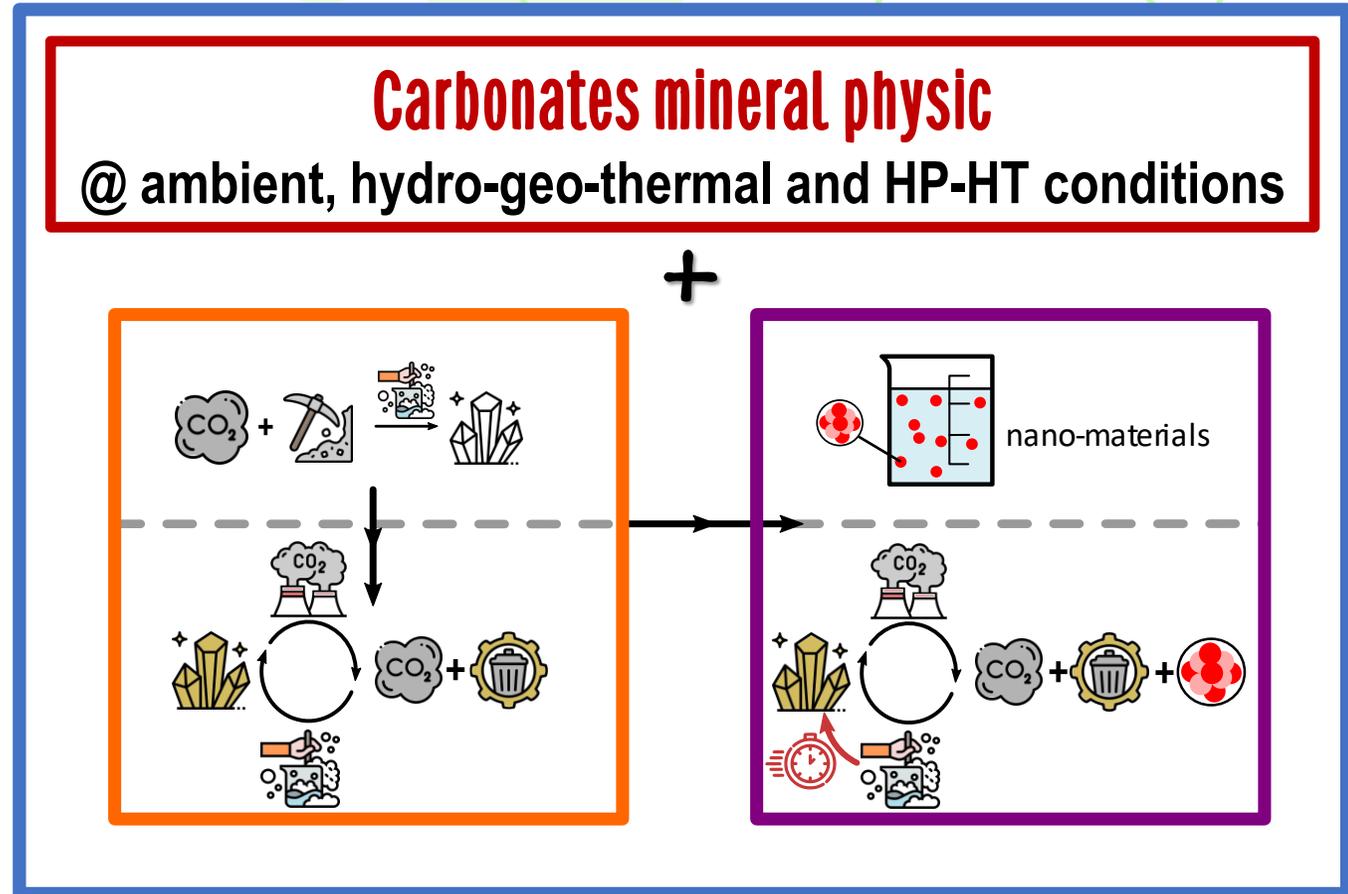
- 1.** Stable carbonate minerals
- 2.** Reagents at **low cost** that will also meet the status of “**end of waste**”
- 3.** Overcoming the slow reaction kinetics

Collaborative actions



Dipartimenti UNIPG:

- **Ingegneria**
- **Scienze Farmaceutiche**
- **Chimica, Biologia e Biotecnologie**
- **Scienze agrarie, alimentari e ambientali**



Thanks for your
attention!!

Azzurra Zucchini, Paola Comodi



A.D. 1308
unipg

UNIVERSITÀ DEGLI STUDI
DI PERUGIA

A.D. 1308
unipg

DIPARTIMENTO
DI FISICA E GEOLOGIA

Dipartimenti:

- Ingegneria
- Scienze Farmaceutiche
- Chimica, Biologia e Biotecnologie
- Scienze agrarie, alimentari e ambientali

In collaboration with:

COLACEM
forte • sostenibile



CIRIAC

Centro Interuniversitario
di Ricerca sull'Inquinamen
da Agenti Fisici - "Mauro F

Partially founded by:



Fondazione Cassa di Risparmio di Perugia



European
Commission

European Regional Development Fund 2014-2020
(ERDF 2014-2020)

European Social Fund 2007-2013 (ESF 2007-2013)

