

# Fisica delle superfici e di film ultrasottili

Dr. Alberto Verdini

CNR-IOM

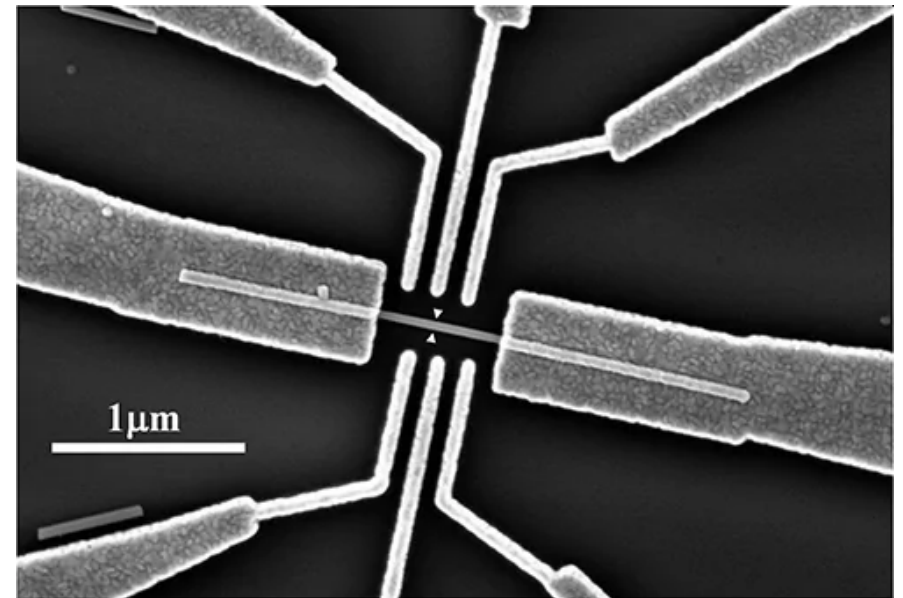
[verdini@iom.cnr.it](mailto:verdini@iom.cnr.it)

# Perché studiare le superfici?



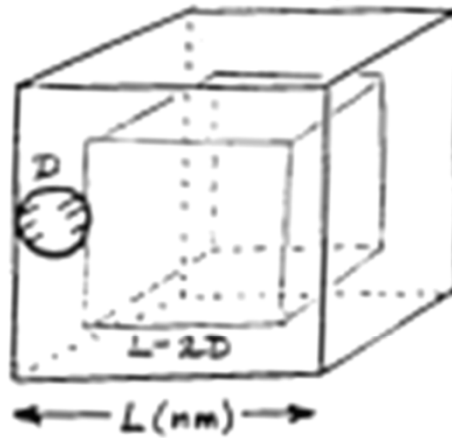
Wafer di Silicio – tecnologia anni '60

## Transistor





FRACTION OF ATOMS ON THE SURFACE  
 OF A CUBE:  $D = \text{ATOMIC DIAM.} = 0.2 \text{ nm} = 2 \text{ \AA}$



**NANOTECHNOLOGIES!**

$$\text{SURFACE FRACTION} = \frac{L^3 - (L - 2D)^3}{L^3}$$

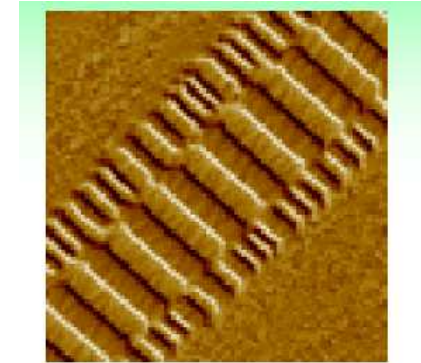
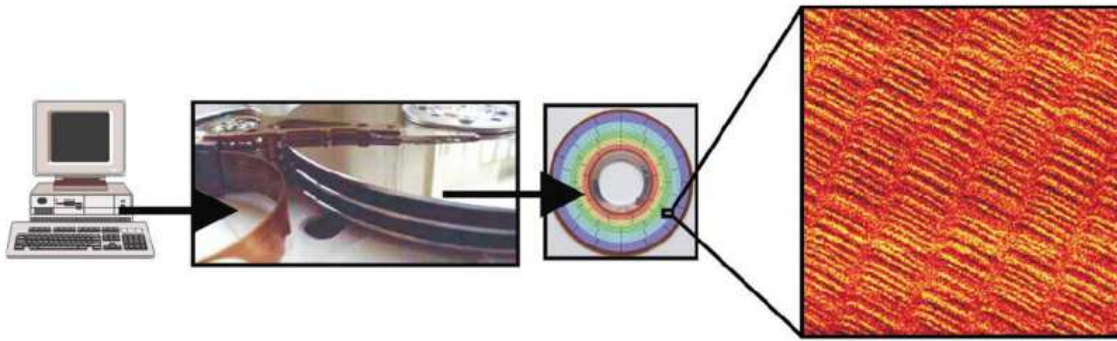
<u>L</u>	<u>FRACTION</u>
$1 \mu\text{m} = 1000 \text{ nm}$	$0.001 = 0.1\%$
$0.1 \mu\text{m} = 100 \text{ nm}$	$0.012 = 1.2\%$
$0.01 \mu\text{m} = 10 \text{ nm}$	$0.115 = 11.5\%$
$0.001 \mu\text{m} = 1 \text{ nm}$	$0.784 = 78.4\%$

*There's Plenty of Room at the Bottom* è il titolo di una presentazione di Richard Feynmann alla riunione dell'APS al Caltech il 29 dicembre 1959.



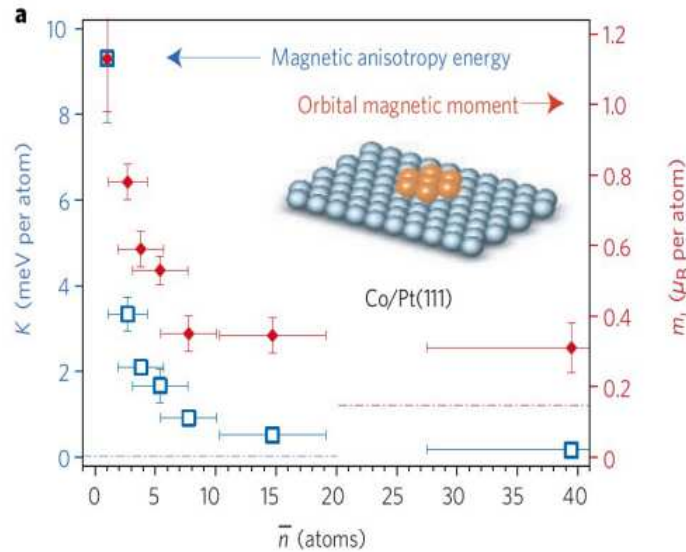
**NANOTECHNOLOGIES !**

# Perché studiare le superfici?



Magnetic hard drive  
(25  $\mu\text{m}$  x 25  $\mu\text{m}$ ).  
Wires are about 2000  
atoms wide.

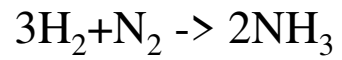
## Data Storage Magnetico



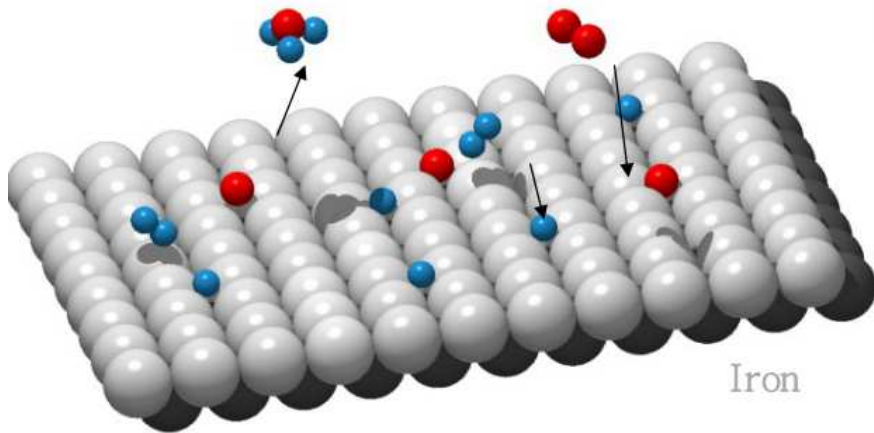
-> Spintronica

# Perché studiare le superfici?

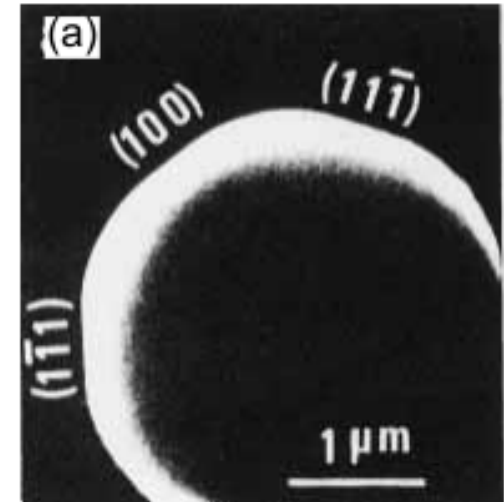
Processo Haber-Bosch (1910)



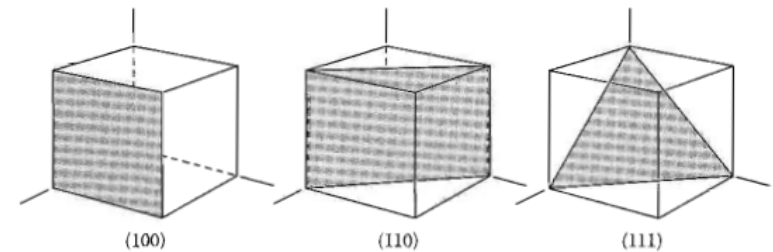
Catalisi eterogenea



Processo utile per fertilizzanti  
usato ancora oggi mediante superfici di Fe

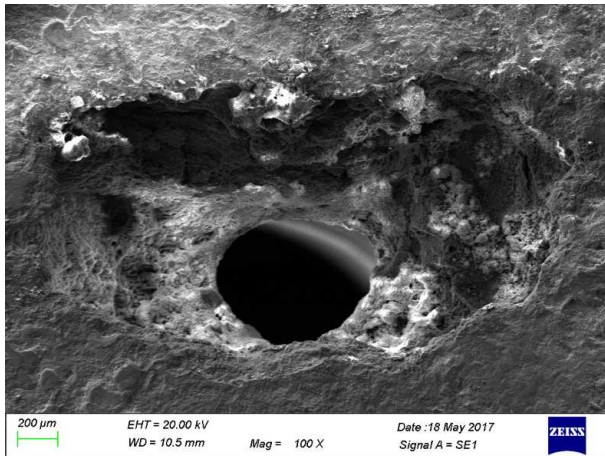


Particella di Pb

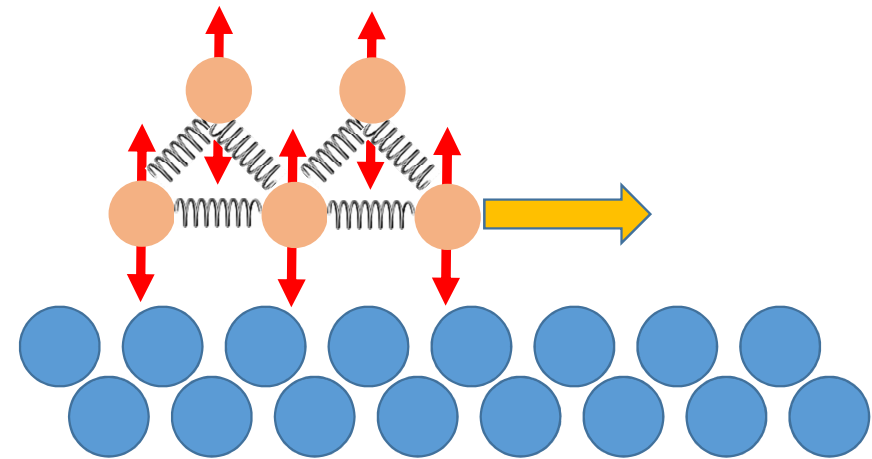
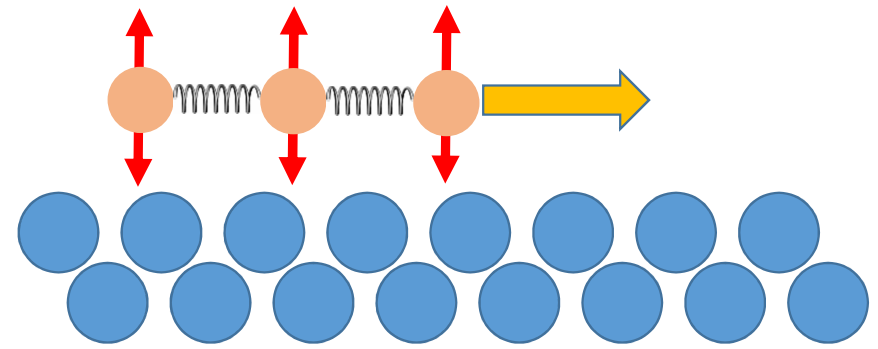


# Perché studiare le superfici?

## Corrosione

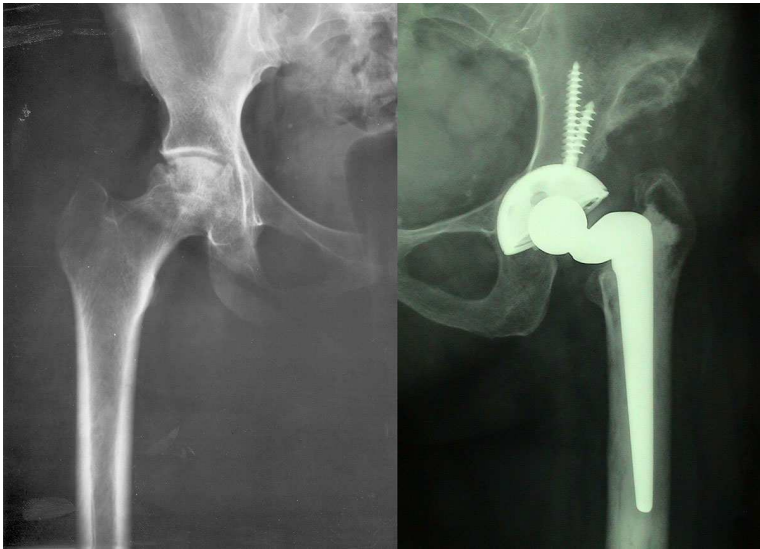


## Attrito, adesione, Tribologia





# Perché studiare le superfici?

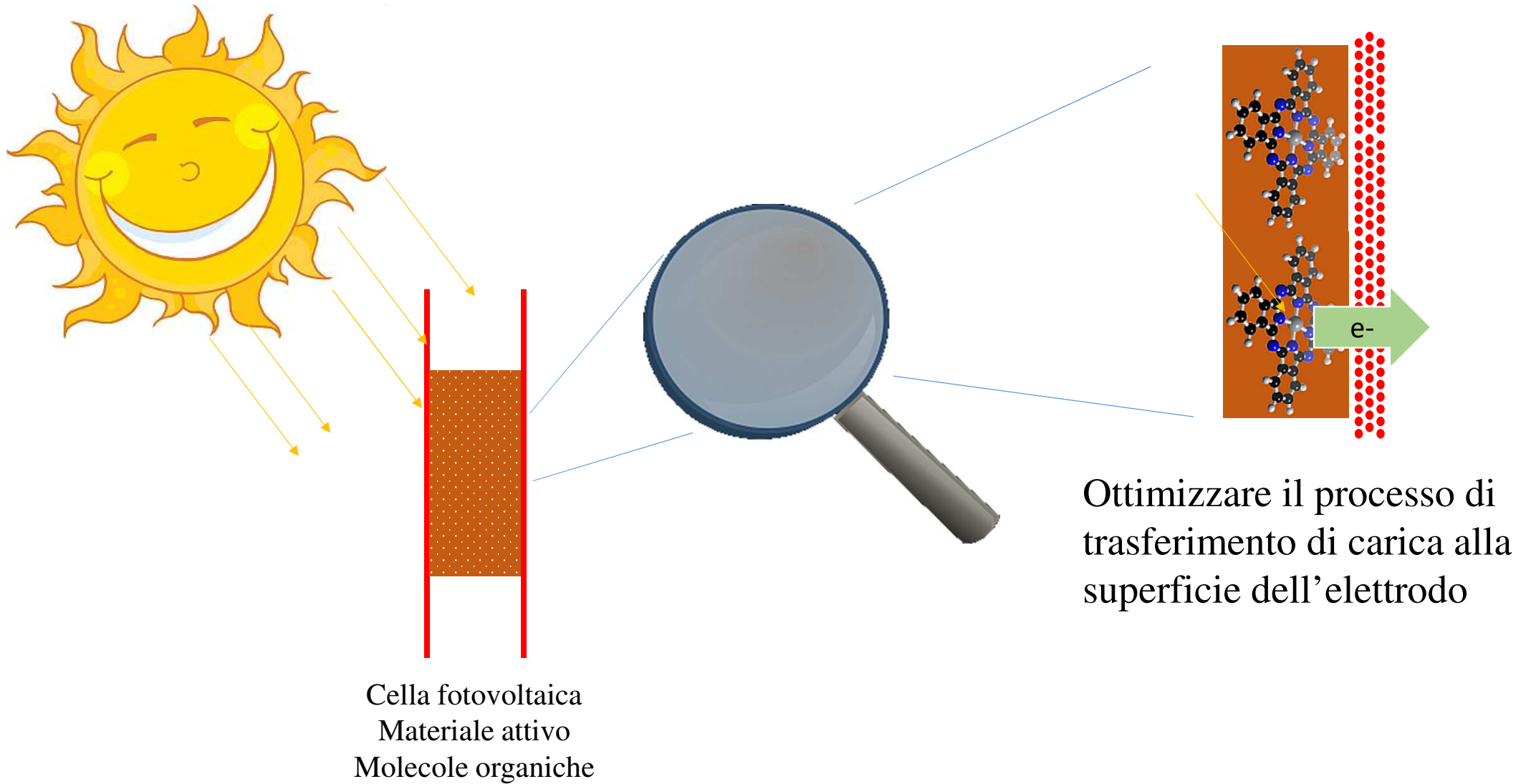


Scienza dei materiali

Medicina

Rivestimento di materiali avanzati in protesi  
(composizione (metalli/polimeri/ceramica), attrito,  
rugosità, tasso di usura, resistenza ai graffi, durezza,  
biocompatibilità, ecc.)

# Perché studiare le superfici?



## Fisica delle Superfici e sistemi modello

- Le superfici reali sono molto complesse e mal definite: policristalli, disordine, difetti, ....
- Dipendono dall'ambiente in cui si trovano
- Impurezze

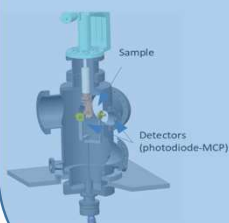
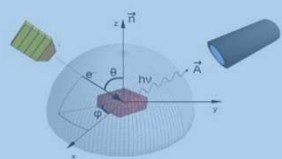
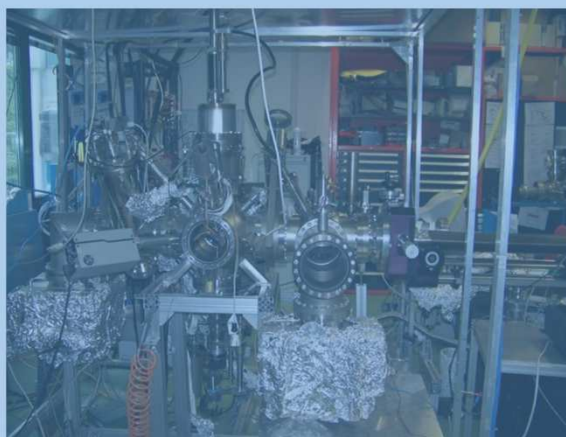
### Come studiarle?

- Partire da sistemi semplici (UHV – ultra alto vuoto)
- Studiare le superfici a «basso indice» di cristalli singoli
- Capire bene queste superfici “ideali” e poi introdurre difetti/irregolarità/disordine in modo controllato
- Rendere i sistemi gradualmente più complessi nella speranza di avere modelli sempre più vicini alla realtà

# Attività di Ricerca @ Laboratorio Congiunto Superfici e Nanostrutture

Dr. Alberto Verdini, Dr.ssa Maddalena Pedio, Prof. Giovanni Carlotti

SIPE-  
nanoStructures  
Inverse  
Photoemission  
and Excitation  
dynamic



ACROSS –Surfaces, Nanostructures,  
Electron Diffraction, Auger and UV  
Photoemission Spectroscopy

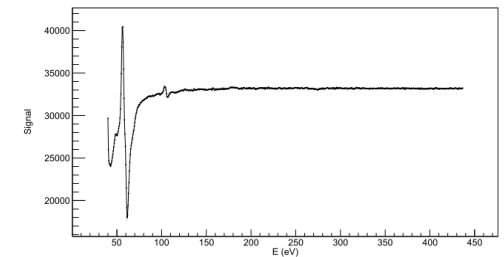
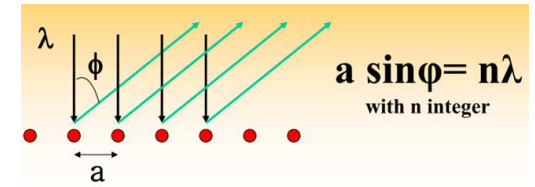


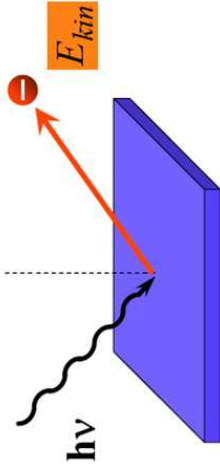
<https://www.iom.cnr.it/research-facilities/facilities-labs/analytical-microscopy-and-spectroscopy/sipe/>

<https://www.iom.cnr.it/research-facilities/facilities-labs/analytical-microscopy-and-spectroscopy/across/>

# Advanced Chamber fOr Surface Studies -> ACROSS

- Low Energy Electron Diffraction - LEED  
+ Auger
- Reflection High Energy Electron  
Diffraction - RHEED
- Electron Bombardment Cell for metals
- Cooled Multicell for organic molecules  
deposition
- Electron Analyzer + 2D detector
- UV Source
- (Monochromatized) X-ray Source

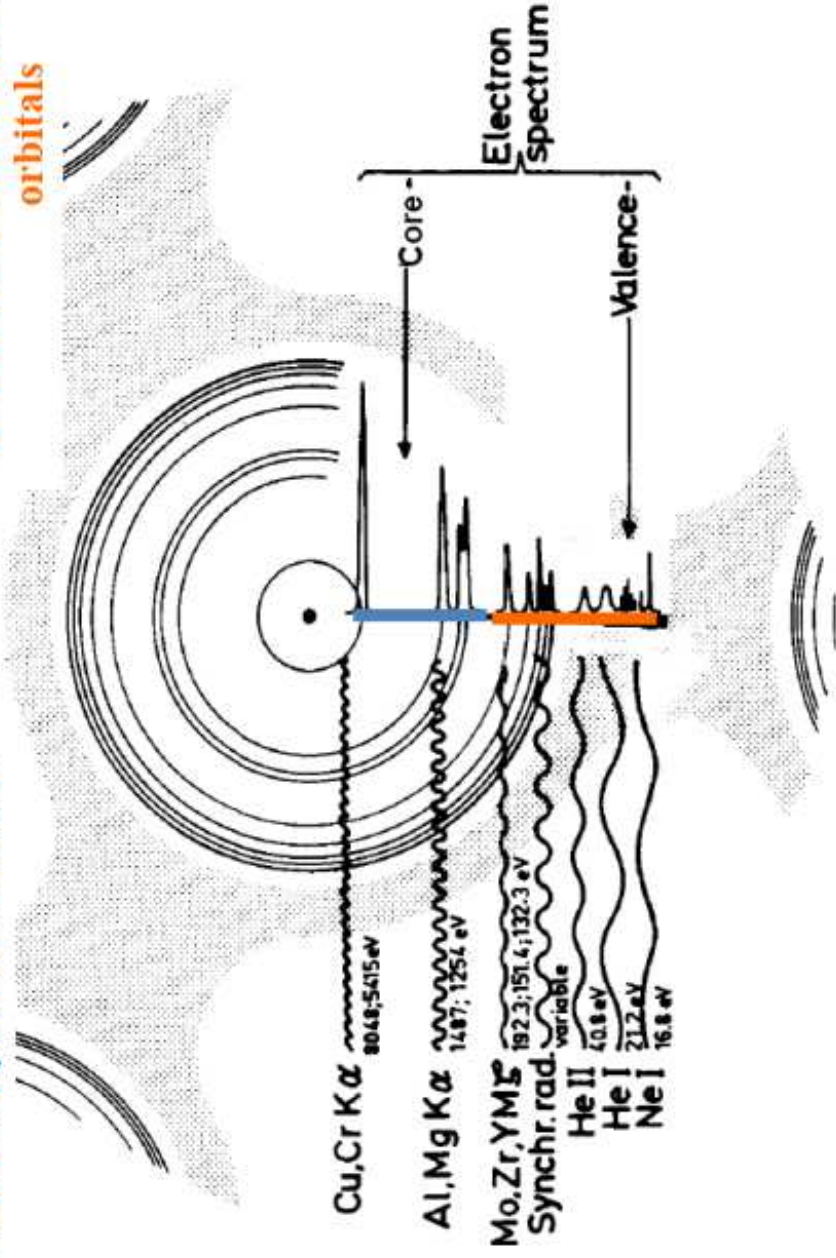




$$E_{kin} = h\nu - E_B - \Phi$$

Localized levels, atomic orbitals

More delocalized levels, molecular orbitals

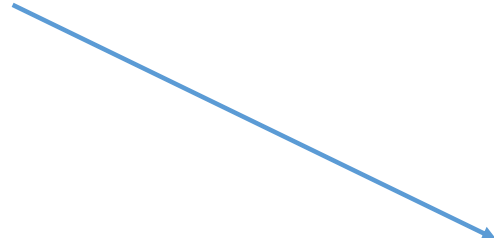
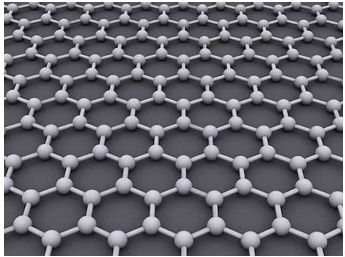




# Attività di Ricerca



# Materiali 2D Monoatomici



Crescita pilotata dalla superficie

Graphene

Silicene

Germanene

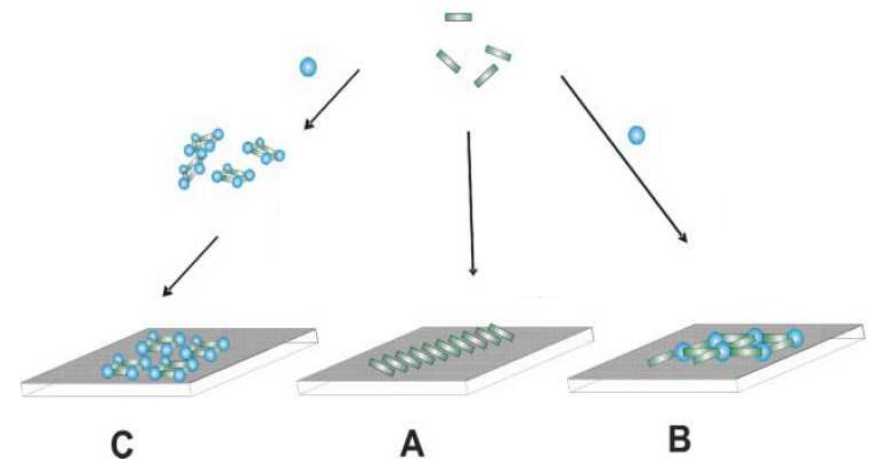
Stanene

Plumbene

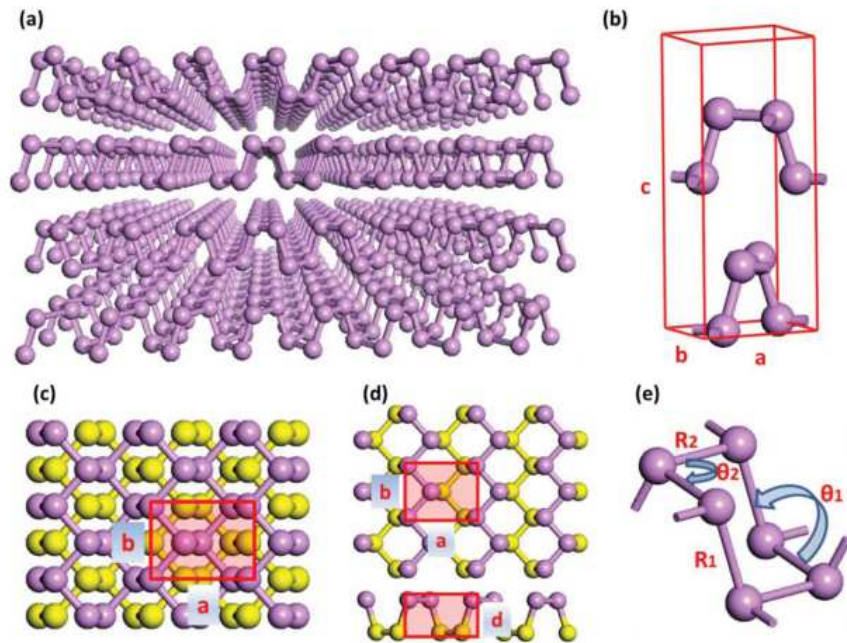
					helium 2 <b>He</b> 4.0026
boron 5 <b>B</b> 10.811	carbon 6 <b>C</b> 12.011	nitrogen 7 <b>N</b> 14.007	oxygen 8 <b>O</b> 15.999	fluorine 9 <b>F</b> 18.998	neon 10 <b>Ne</b> 20.180
aluminum 13 <b>Al</b> 26.982	silicon 14 <b>Si</b> 28.086	phosphorus 15 <b>P</b> 30.974	sulfur 16 <b>S</b> 32.065	chlorine 17 <b>Cl</b> 35.453	argon 18 <b>Ar</b> 39.948
gallium 31 <b>Ga</b> 69.723	germanium 32 <b>Ge</b> 72.61	arsenic 33 <b>As</b> 74.922	selenium 34 <b>Se</b> 78.96	bromine 35 <b>Br</b> 79.904	krypton 36 <b>Kr</b> 83.80
indium 49 <b>In</b> 114.82	tin 50 <b>Sn</b> 118.71	antimony 51 <b>Sb</b> 121.76	tellurium 52 <b>Te</b> 127.60	iodine 53 <b>I</b> 126.90	xenon 54 <b>Xe</b> 131.29
thallium 81 <b>Tl</b> 204.38	lead 82 <b>Pb</b> 207.2	bismuth 83 <b>Bi</b> 208.98	polonium 84 <b>Po</b> [209]	astatine 85 <b>At</b> [210]	radon 86 <b>Rn</b> [222]

Borophene

Phosphorene



# Black Phosphorous



PHYSICAL REVIEW B **93**, 035448 (2016)

## Surface structure determination of black phosphorus using photoelectron diffraction

Luis Henrique de Lima,<sup>1,\*</sup> Lucas Barreto,<sup>1,2</sup> Richard Landers,<sup>1</sup> and Abner de Siervo<sup>1,†</sup>

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(Received 16 November 2015; revised manuscript received 11 January 2016; published 26 January 2016)

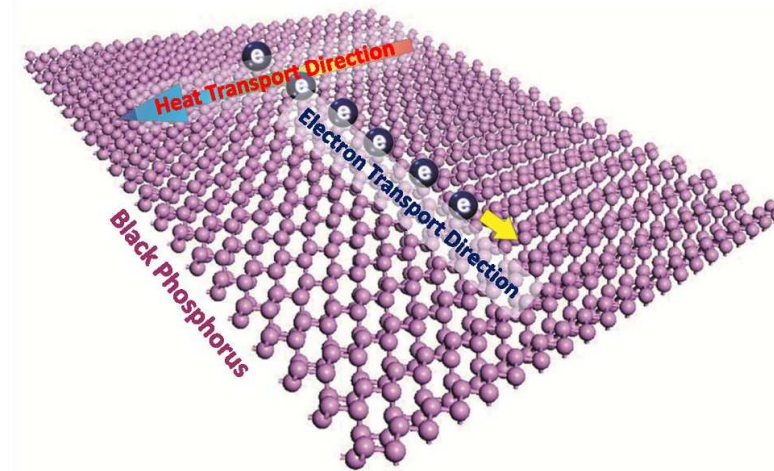
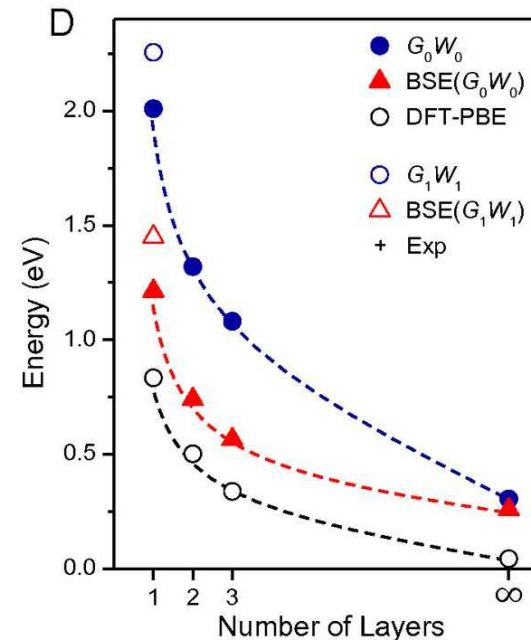
## Applicazioni

- Transistor FET
- Dispositivi Optoelettronici
- Celle solari
- Scissione fotocatalitica dell'acqua
- Batterie Li-ion
- Materiali termoelettrici
- Sensori

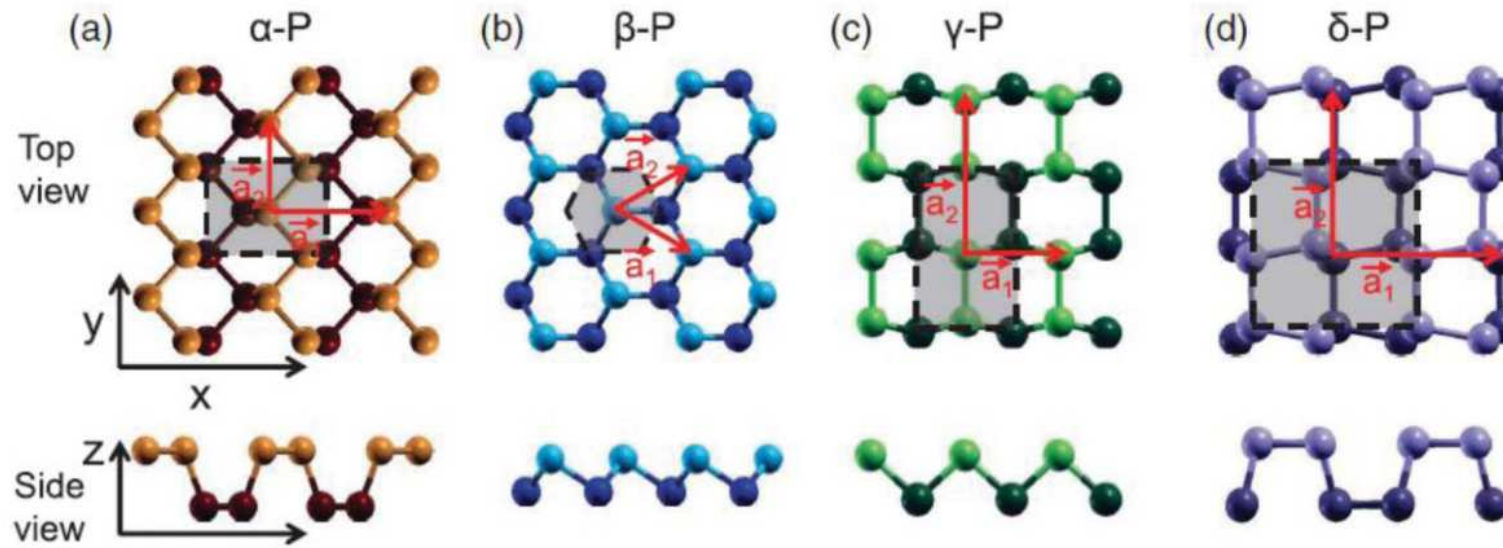
V. Sorkin et al. Critical Reviews in Solid State and Materials Sciences, 42, 1 (2016)

# Black phosphorous principali proprietà interessanti:

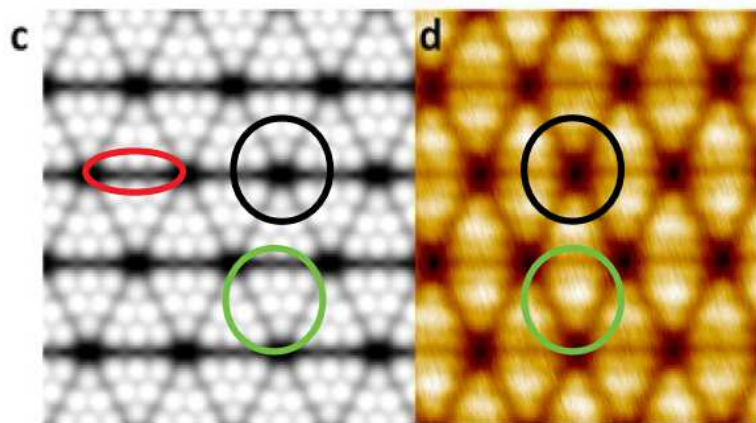
- La gap dipende dal numero di strati
- Anisotropia sul piano del trasporto di calore e di elettroni



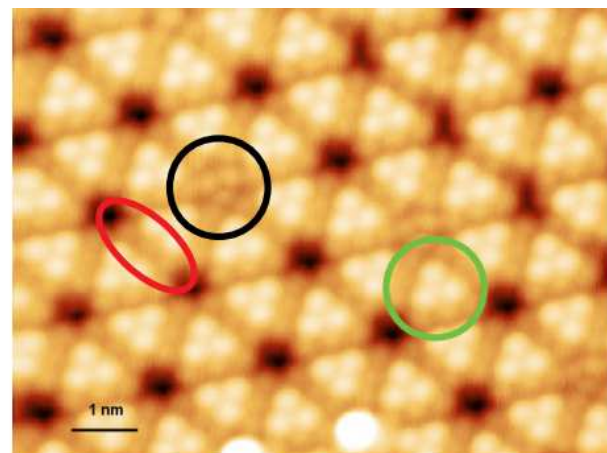
# Molti possibili monostrati con diverse strutture



Fasi calcolate per un singolo strato di P

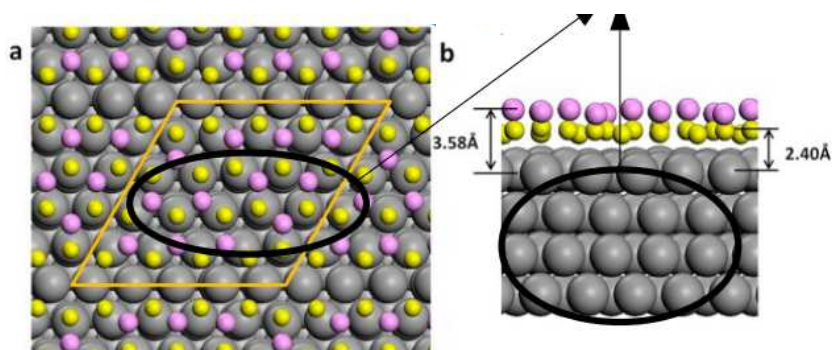


[1]J.L. Zhang et al., Epitaxial Growth of Single Layer Blue Phosphorus: A New Phase of Two-Dimensional Phosphorus, Nano Lett. 16, 4903-4908 (2016)

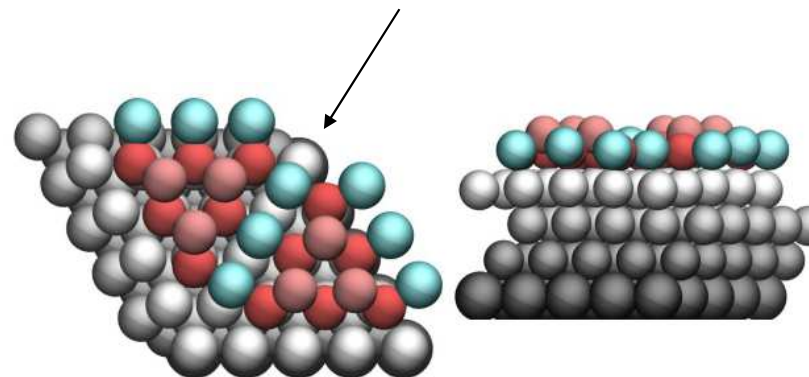


Our expSTM image -  $P_{16} \times 2$  model

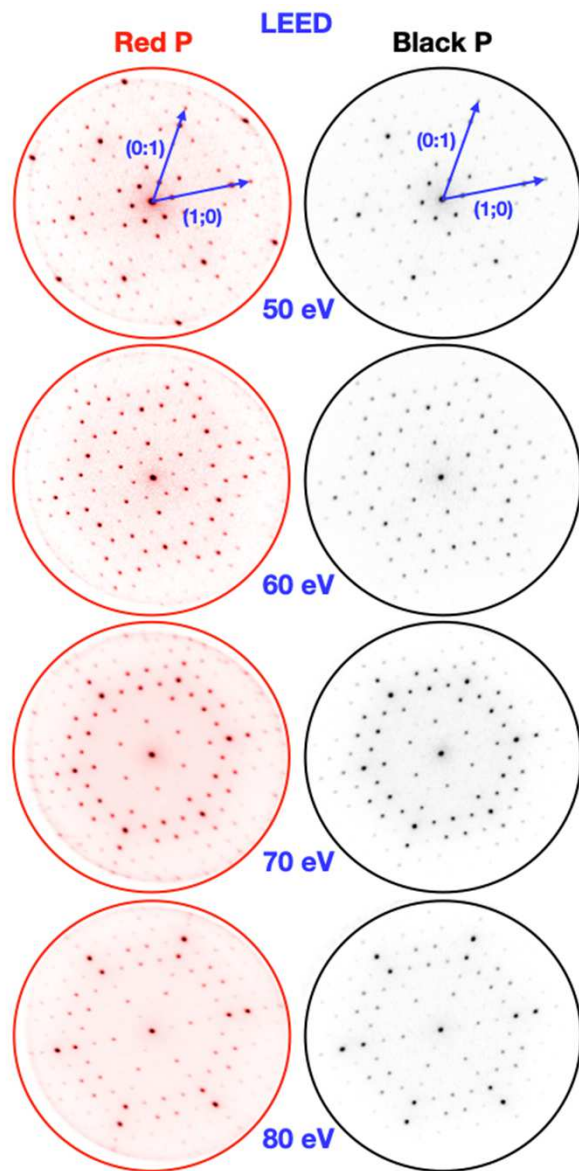
Struttura non stabile



? P o Au ?

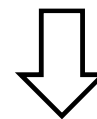


# Electron Diffraction

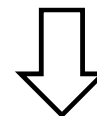


Perché non usare il fosforo rosso (molto molto economico) invece di quello nero (molto costoso)?

Stesse Strutture



SI!



Abbassamento dei costi (anche ambientali) per la produzione di dispositivi

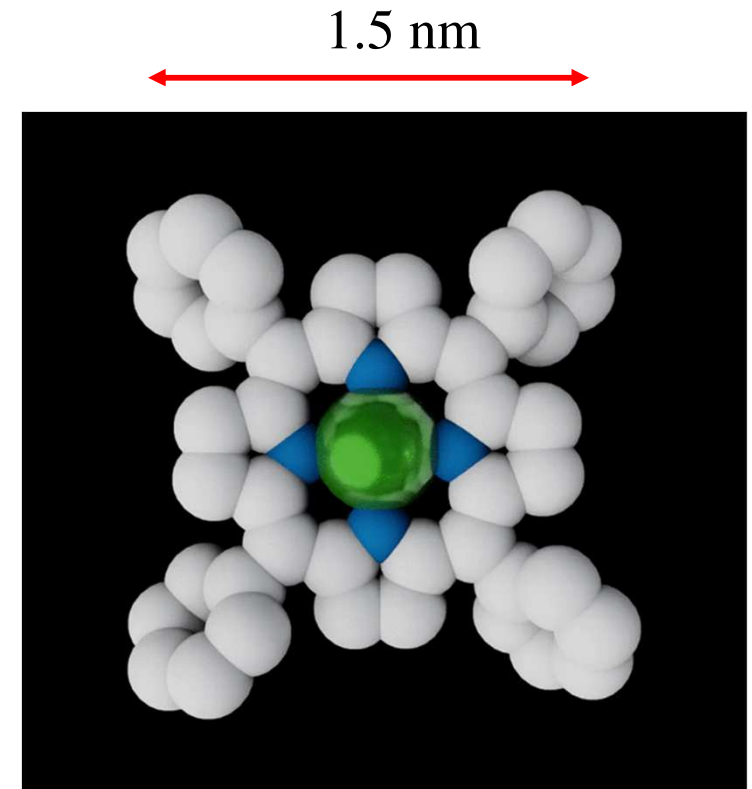
# Porfirina (tetrapirrolo), molecola metallorganica -> semiconduttori organici

Funzionalizzabile mediante diversi ligandi

Metallo al centro: Cu, Zn, Ni, Fe, Pd, ....

Applicazioni:

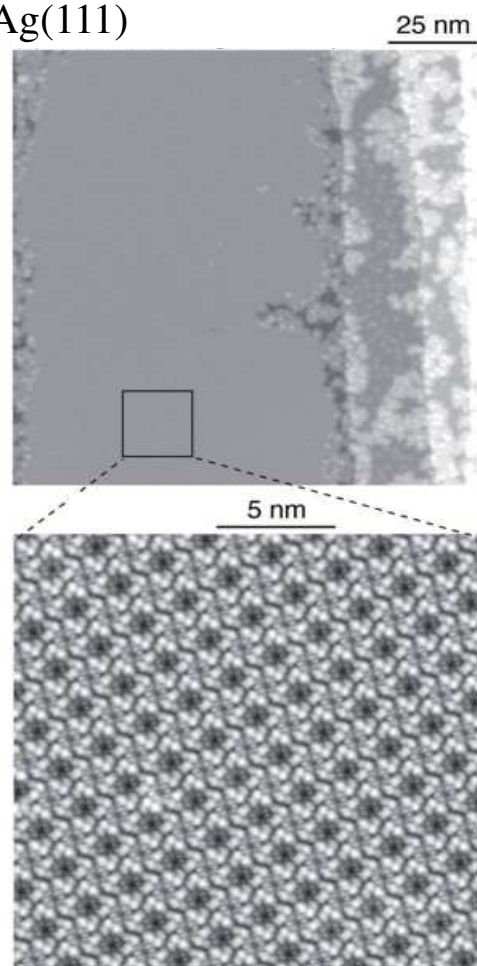
- Celle Solari
- Dispositivi elettronici a basso consumo
- Transistor molecolari
- Spintronica
- H<sub>2</sub> da Water Splitting
- Sensori
- Quantum Computing
- Terapie anti tumorali



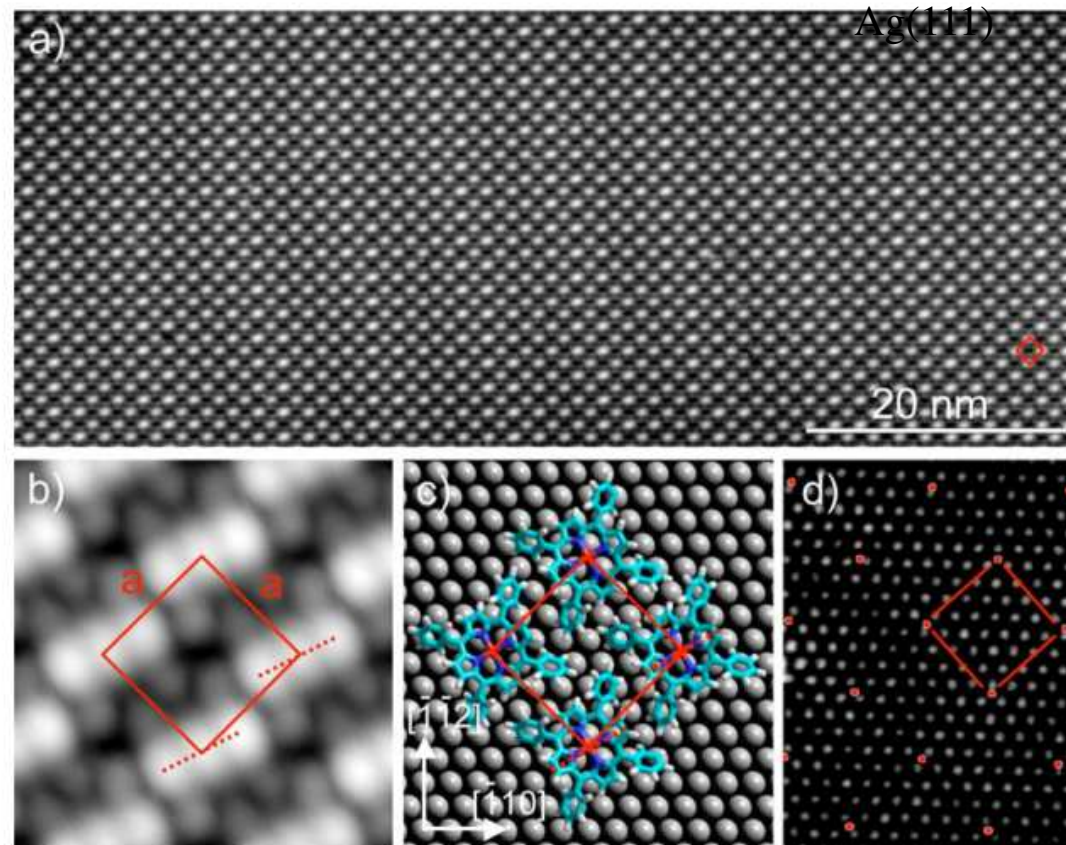
Presente nei sistemi biologici fondamentali  
Emoglobina, Clorofilla, F430,...

# Autoassemblamento sulle superfici

TbPc<sub>2</sub>/ Ag(111)



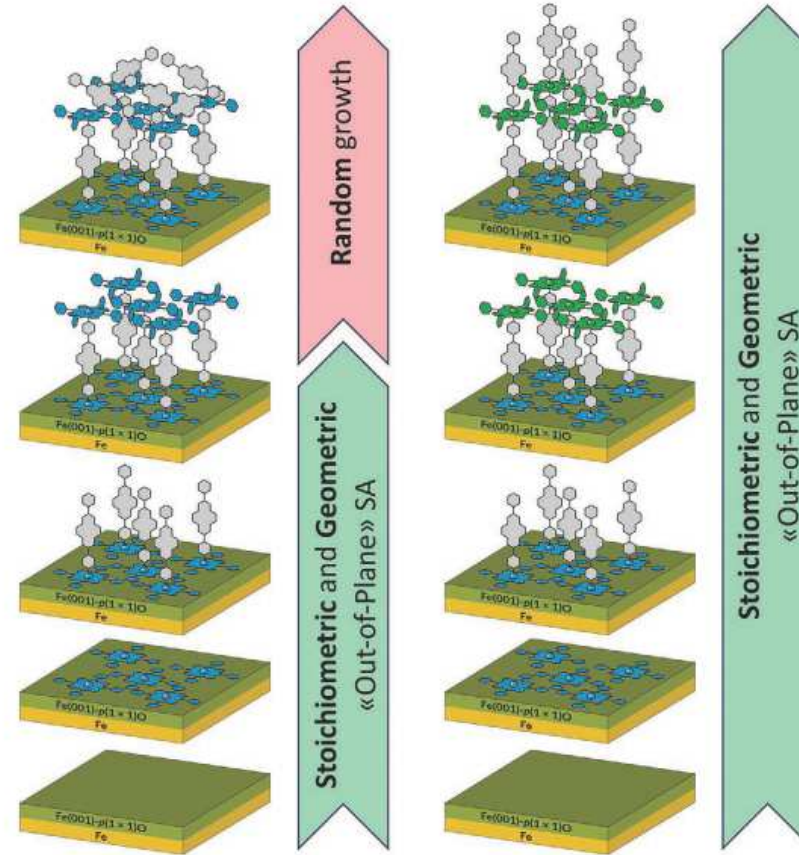
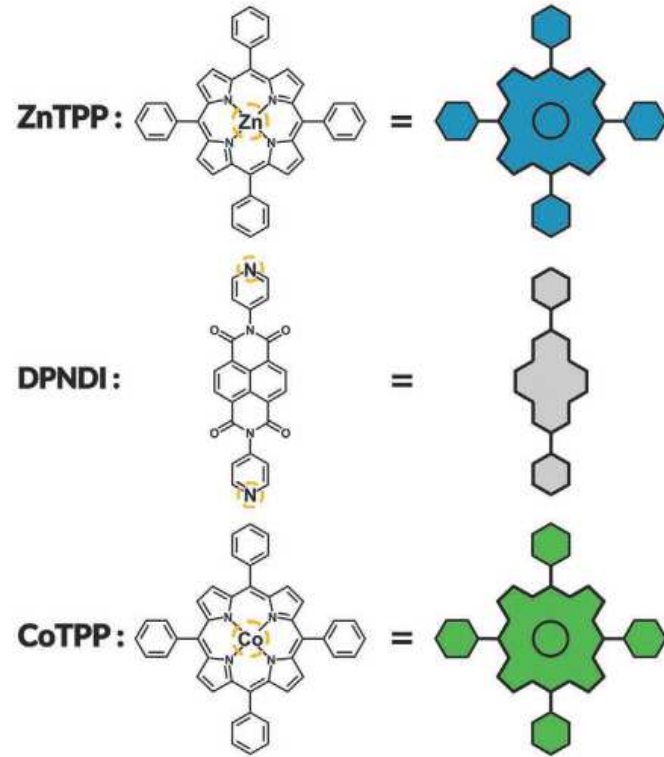
CoTPP/  
Ag(111)





# Autoassemblamento in 3D

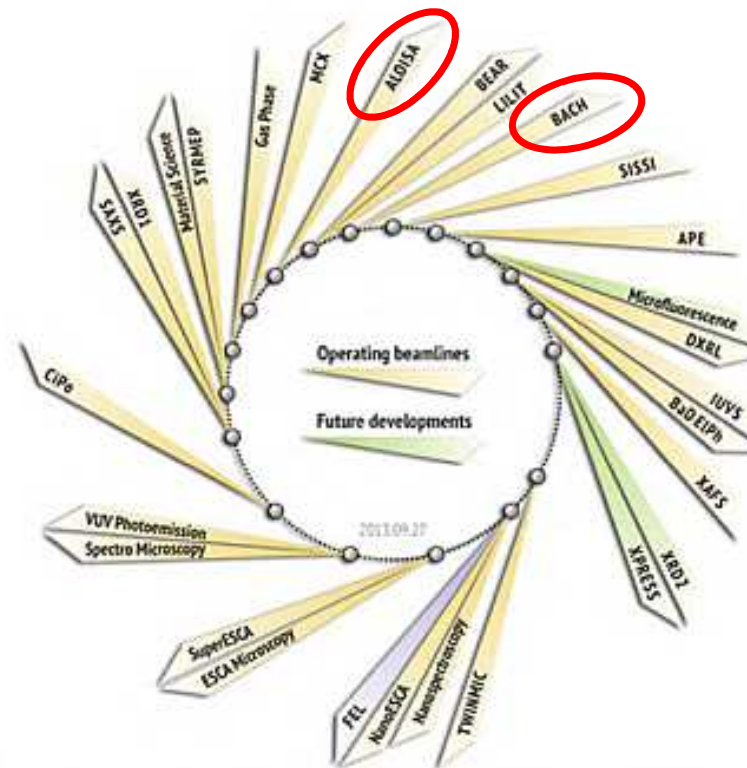
coordination site 



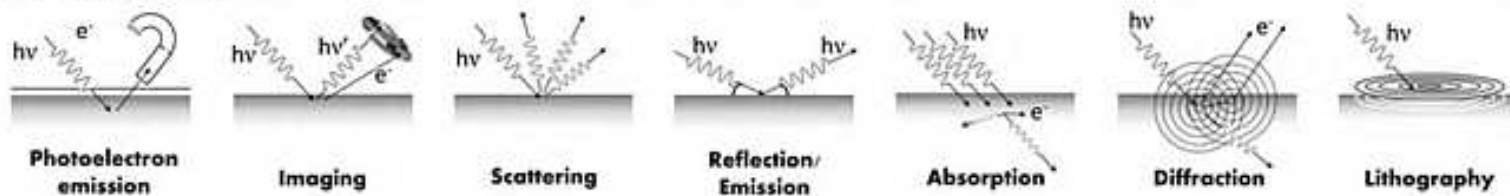
A. Bossi, A. Orbelli Biroli @CNR-SCITEC MI  
 G. Bussetti et al. @Polimi  
 L. Floreano, L. Schio @CNR-IOM TS  
 Adv. Funct. Mater. 2021, 31, 2011008

# Attività presso Sincrotrone Elettra - Trieste

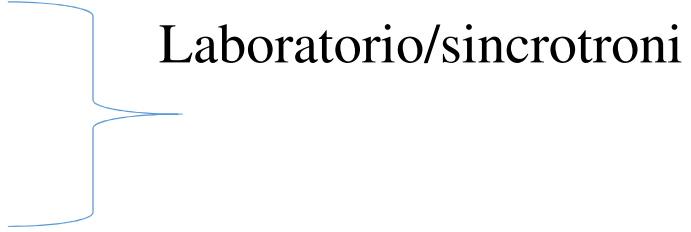
## Elettra



Beamlines by technique



## Possibili argomenti per le tesi triennali e magistrali

- Studio della crescita di P su superfici metalliche:  
Cu(111), Cu(100), Ni(111), Pt(111),....
  - Studio delle interazioni monostrato di P con tetrapirroli
  - Ottimizzazione della tecnica di spettroscopia UPS e XPS
  - Ottimizzazione della strumentazione delle camere sperimentali –  
motorizzazioni, acquisizione dati, automazione,...
- 
- Laboratorio/sincrotroni

**Grazie per la vostra attenzione**